

**MATHEMATICS**

1. C

**Sol.** Doubtful points :  $-1, 0, 1, \sqrt{2}, \sqrt{3}, 2$

at  $x = \sqrt{2}, \sqrt{3}$

$$f(x) = (2x^2 + \underset{\text{Cont.}}{x} - [x]) + [\underset{\text{Cont.}}{x^2}] = \text{Discount}$$

at  $x = -1$  ;

$$\text{RHL} \Rightarrow \left. \begin{aligned} f(x) &= (2 - 1 - (-1)) + 0 = 2 \\ f(-1) &= 2 - 1 - (-1) + 1 = 3 \end{aligned} \right\} \text{Dis.}$$

at  $x = 2$

$$\text{LHL} \Rightarrow \left. \begin{aligned} f(x) &= 8 + 2 - 1 + 3 = 12 \\ f(2) &= 8 + 2 - 2 + 4 = 12 \end{aligned} \right\} \text{Cont.}$$

at  $x = 0$

$$\text{LHL} \Rightarrow \left. \begin{aligned} 0 + 0 - (-1) + 0 &= 1 \\ f(=0) \end{aligned} \right\} \text{Cont.}$$

at  $x = 1$

$$\left. \begin{aligned} \text{LHL} \Rightarrow 2 + 1 - 0 + 0 &= 3 \\ f(1) &= 3 - 1 + 1 = 3 \\ \text{RHL} \Rightarrow 2 + 1 - 1 + 1 &= 3 \end{aligned} \right\} \text{Cont.}$$

2. C

**Sol.**  $C \equiv x^2 + y^2 + gx + gy = 0 \quad \dots(1)$

$$2x + 2yy' + g + gy' = 0$$

$$g = -\left(\frac{2x + 2yy'}{1 + y'}\right)$$

Put in (1)

$$x^2 + y^2 - \left(\frac{2x + 2yy'}{1 + y'}\right)(x + y) = 0$$

$$(x^2 - y^2 - 2xy)y' = x^2 - y^2 + 2xy$$

3. D

**Sol.**  $S_1 : x^2 + y^2 \leq 25 \quad \dots(1)$

$$S_2 : \text{Im of } \frac{z + (1 - \sqrt{3}i)}{(1 - \sqrt{3}i)} \geq 0$$

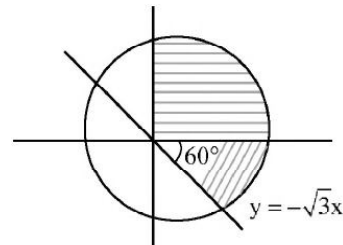
$$\text{Im of } \left(\frac{x + iy}{1 - \sqrt{3}i} + 1\right) \geq 0$$

$$\text{Im of } \left(\frac{(x + y)(1 + \sqrt{3}i)}{4}\right) \geq 0$$

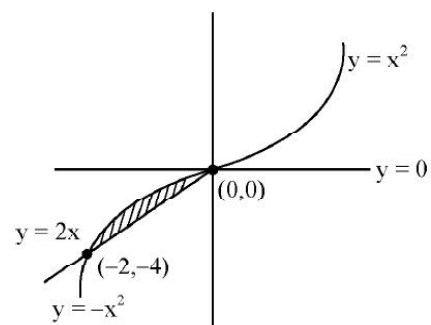
$$\Rightarrow \sqrt{3}x + y \geq 0 \quad \dots(2)$$

$$S_3 : x \geq 0 \quad \dots(3)$$

$$\text{Area} = \frac{5}{12}(\pi(5)^2)$$



4. D



$$A = \int_{-2}^0 -x^2 - 2x = \frac{4}{3}$$

5. C

**Sol.** B B H J O

**B** \_\_\_\_\_  $4! = 24$

**H** \_\_\_\_\_  $\frac{4!}{2!} = 12$

**J** \_\_\_\_\_  $\frac{4!}{2!} = 12$

O B B H J

O B B J H  $\rightarrow$  50<sup>th</sup> rank

6. B

**Sol.** Let's assume  $\vec{v} = \vec{a} + \vec{b} + \hat{i}$

$$= 5\hat{i} + 3\hat{j} + \hat{k}$$

and  $\vec{c} = \hat{i} = \vec{p}$

So,

$$\vec{p} \times \vec{v} = \vec{a} \times \vec{p}$$

$$\vec{p} \times \vec{v} + \vec{p} \times \vec{a} = \vec{0}$$

$$\vec{p} \times (\vec{v} + \vec{a}) = \vec{0}$$

$$\Rightarrow \vec{p} = \lambda(\vec{v} + \vec{a})$$

$$\vec{c} + \hat{i} = \lambda(7\hat{i} + 8\hat{j})$$

$$\vec{a} \cdot \vec{c} + \vec{a} \cdot \hat{i} = \lambda \vec{a} \cdot (7\hat{i} + 8\hat{j})$$

$$-29 + 2 = \lambda(14 + 40)$$

$$\lambda = -\frac{1}{2}$$

$$\vec{c} \cdot (-2\hat{i} + \hat{j} + \hat{k}) + \hat{i} \cdot (-2\hat{i} + \hat{j} + \hat{k})$$

$$= \lambda(7\hat{i} + 8\hat{j}) \cdot (-2\hat{i} + \hat{j} + \hat{k})$$

$$= -\frac{1}{2}(-14 + 8) + 2 = 5$$

7. C

**Sol.**  $|\vec{c} - \vec{a}| = |\vec{c}|^2 + |\vec{a}|^2 - 2\vec{a} \cdot \vec{c}$

$$= |\vec{c}|^2 + 4 - 0$$

$$\therefore \vec{a} = \vec{b} \times \vec{c}$$

$$|\vec{a}| = |\vec{b} \times \vec{c}|$$

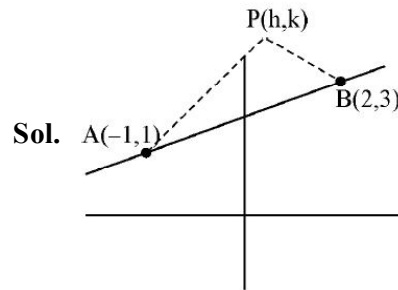
$$2 = 3|\vec{c}| \sin \alpha$$

$$\alpha \in \left[ 0, \frac{\pi}{3} \right]$$

$$|\vec{c}|_{\min} = \frac{2}{3} \times \frac{2}{\sqrt{3}} \quad \text{cosec} \alpha \in \left[ \frac{2}{\sqrt{3}}, \infty \right)$$

$$\Rightarrow 27|\vec{c} - \vec{a}|_{\min}^2 = 27 \left( \frac{16}{27} + 4 \right) = 124$$

8. A



$$\frac{1}{2} \begin{vmatrix} h & k & 1 \\ -1 & 1 & 1 \\ 2 & 3 & 1 \end{vmatrix} = 10$$

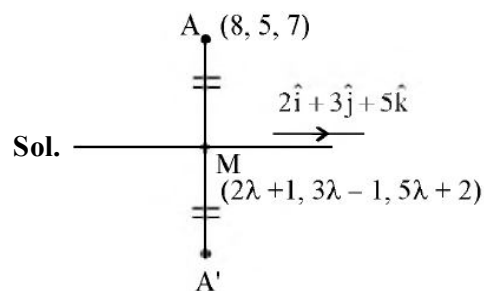
$$-2x + 3y = 25$$

$$-\frac{6}{5}x + \frac{9}{5}y = 15$$

$$a = -\frac{6}{5}, b = \frac{9}{5}$$

$$5a = -6, 2b = \frac{18}{5}$$

9. C



$$\overline{AM} \cdot (2\hat{i} + 3\hat{j} + 5\hat{k}) = 0$$

$$(2\lambda - 7)(2) + (3\lambda - 6)(3) + (5\lambda - 5)(5) = 0$$

$$38\lambda = 57$$

$$\lambda = \frac{3}{2}$$

$$M \left( 4, \frac{7}{2}, \frac{19}{2} \right)$$

$$A'(0, 2, 12)$$

10. C

**Sol.**  $T_{r+1} = {}^{12}C_r \left(\frac{3^{1/5}}{x}\right)^{12-r} \left(\frac{2x}{5^{1/3}}\right)^r$

$$T_{r+1} = \frac{{}^{12}C_r (3)^{\frac{12-r}{5}} (2)^r (x)^{2r-12}}{(5)^{r/3}}$$

$r = 6$

$$T_7 = \frac{{}^{12}C_6 (3)^{6/5} (2)^6}{5^2} = \left(\frac{9 \times 11 \times 7}{25}\right) 2^8 \cdot 3^{1/5}$$

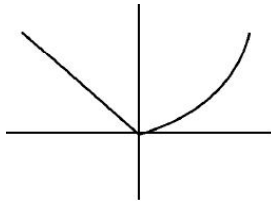
$25\alpha = 693$

11. A

**Sol.**  $f(g(x)) = |g(x) - 1|$

$$\text{fog} \begin{cases} |e^x - 1| & x \geq 0 \\ |x + 1 - 1| & x \leq 0 \end{cases}$$

$$\text{fog} \begin{cases} e^x - 1 & x \geq 0 \\ -x & x \leq 0 \end{cases}$$



12. A

**Sol.**  $S_1 : x^2 + y^2 - 2x - 2y + 1 = 0$

$S_2 : x^2 + y^2 + 2x - 3 = 0$

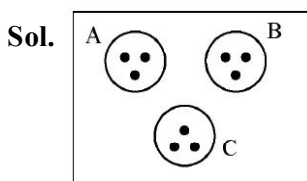
Common chord =  $S_1 - S_2 = 0$

$-4x - 2y + 4 = 0$

$2x + y = 2 \Rightarrow P(0, 2)$

$d_{(c,p)}^2 = (1-0)^2 + (2-1)^2 = 2$

13. A



$$\frac{9!}{(3!3!3!)} \times 3!$$

14. D

**Sol.**  $D = \begin{vmatrix} 1 & 1 & 1 \\ 2 & 5 & 5 \\ 1 & 2 & m \end{vmatrix} = 0 \Rightarrow m = 2$

$$D_3 = \begin{vmatrix} 1 & 1 & 4 \\ 2 & 5 & 17 \\ 1 & 2 & n \end{vmatrix} = 0 \Rightarrow n = 7$$

15. B

**Sol.**  $D > 0$

$b^2 > 4ac$

$b = 3 : (a, c) = (1, 1)(1, 2)(2, 1)$

$b = 4 : (a, c) = (1, 1)(1, 2)(2, 1)(1, 3)(3, 1)$

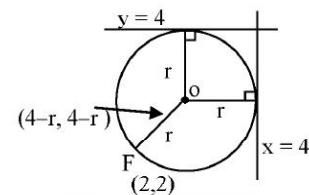
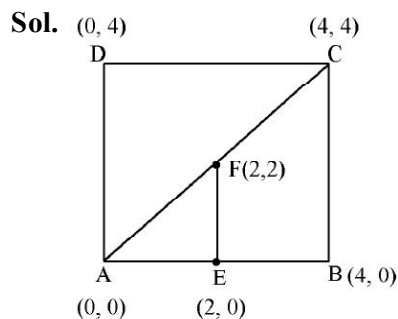
$b = 5 : (a, c) = (1, 1)(1, 2)(2, 1)(1, 3)(3, 1)(1, 4)(4, 1)(1, 5)(5, 1)(1, 6)(6, 1)(2, 3)(3, 2)(2, 2)$

$b = 6 : (a, c) = (1, 1)(1, 2)(2, 1)(1, 3)(3, 1)(1, 4)(4, 1)(1, 5)(5, 1)(1, 6)(6, 1)(2, 3)(3, 2)(2, 4)(4, 2)(2, 2)$

fav. cases = 38

Prob. :  $\frac{38}{6 \times 6 \times 6}$

16. B



$OF^2 = r^2$

$(2-r)^2 + (2-r)^2 = r^2$

$r^2 - 8r + 8 = 0$

17. D

**Sol.**  $I = \int_0^1 1 \cdot (1-x^{10})^{20} dx$

$$x^{10} = t$$

$$x = t^{1/10}$$

$$dx = \frac{1}{10}(t)^{-9/10} dt$$

$$I = \int_0^1 (1-t)^{20} \frac{1}{10}(t)^{-9/10} dt$$

$$I = \frac{1}{10} \int_0^1 t^{-9/10} (1-t)^{20} dt$$

$$a = \frac{1}{10} \quad b = \frac{1}{10} \quad c = 21$$

18. D

Equation co-factor of  $A_{21}$

$$(2\alpha^2 - 3\alpha) = \alpha$$

$$\alpha = 0, 2 \text{ (accept)}$$

$$\text{Now, } 2\alpha^2 - \alpha\beta = 3\alpha$$

$$\alpha = 2 \quad \beta = 1$$

$$|AB| = |A \text{ cof}(A)| = |A|^3$$

$$A = \begin{vmatrix} 1 & 2 & 3 \\ 2 & 2 & 1 \\ -1 & 2 & 4 \end{vmatrix} = 6 - 2(9) + 3(6) = 6$$

19. D

$$\text{Sol. } y = \frac{2 \cos \theta + 2 \cos^2 \theta - 1}{4 \cos^3 \theta - 3 \cos \theta + 8 \cos^2 \theta - 4 + 5 \cos \theta + 2}$$

$$y = \frac{(2 \cos^2 \theta + 2 \cos \theta - 1)}{(2 \cos^2 \theta + 2 \cos \theta - 1)(2 \cos \theta + 2)}$$

$$y = \frac{1}{2} \left( \frac{1}{1 + \cos \theta} \right)$$

$$\Rightarrow \theta = \frac{\pi}{2} \quad y = \frac{1}{2}$$

$$y'' = \frac{1}{2} \left[ \frac{\cos \theta (1 + \cos \theta)^2 - \sin \theta (2)(1 + \cos \theta)(-\sin \theta)}{(1 + \cos \theta)^4} \right]$$

$$\Rightarrow \theta = \frac{\pi}{2} \quad y = 1$$

20. A

$$\text{Sol. } k = 4 \left( 4^x + \frac{1}{4^x} \right) + \left( 4^{2x} + \frac{1}{4^{2x}} \right) \\ \geq 2 \quad \geq 2$$

$$k \geq 10$$

21. 5

$$\text{Sol. } \frac{1}{3} + k + \frac{1}{6} + \frac{1}{4} = 1 \quad \Rightarrow k = \frac{1}{4}$$

$$\mu = \frac{\alpha}{3} + \frac{1}{4} - \frac{3}{4}$$

$$\mu = \frac{\alpha}{3} - \frac{1}{2}$$

$$\sigma = \sqrt{\left( \alpha^2 + \frac{1}{3} + \frac{1}{4} + 9 \frac{1}{4} \right) - \left( \frac{\alpha}{3} - \frac{1}{2} \right)^2}$$

$$\sigma = \sqrt{\frac{2\alpha^2}{9} + \frac{\alpha}{3} + \frac{9}{4}}$$

$$\sigma = \mu + 2$$

$$\sigma = (\mu + 2)^2 \Rightarrow \frac{2\alpha^2}{9} + \frac{\alpha}{3} + \frac{9}{4} = \frac{\alpha^2}{9} + \frac{9}{4} + \alpha$$

$$\frac{\alpha^2}{9} - \frac{2\alpha}{3} = 0$$

$$\alpha = 0, \text{ (reject) or } \alpha = 6$$

( $\because x = 0$  is already given)

$$\Rightarrow \alpha + \mu = 2\mu + 2$$

$$= 5$$

22. 18

$$\text{Sol. IF} = e^{\int \frac{2x}{(1+x^2)^2} dx} = e^{\frac{-1}{1+x^2}}$$

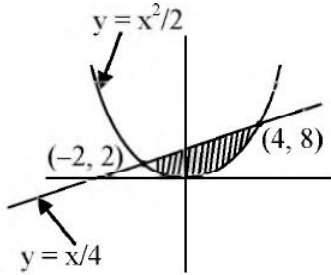
$$y \cdot e^{\frac{-1}{1+x^2}} = \int x \cdot e^{\frac{1}{1+x^2}} \cdot e^{\frac{-1}{1+x^2}} dx$$

$$y \cdot e^{\frac{-1}{1+x^2}} = \frac{x^2}{2} + c$$

$$(0, 0) \Rightarrow \boxed{C = 0}$$

$$y(x) = \frac{x^2}{2} e^{\frac{1}{1+x^2}}$$

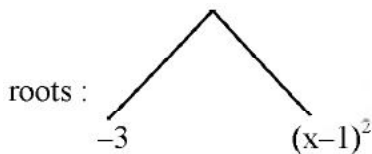
$$f(x) = \frac{x^2}{2}$$



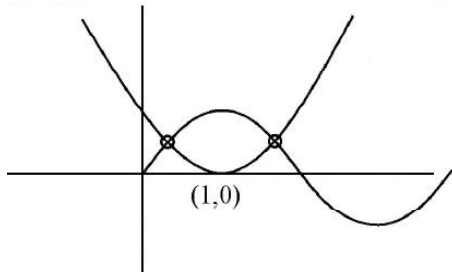
$$A = \int_{-2}^4 (x+4) - \frac{x^2}{2} dx = 18$$

23. 2

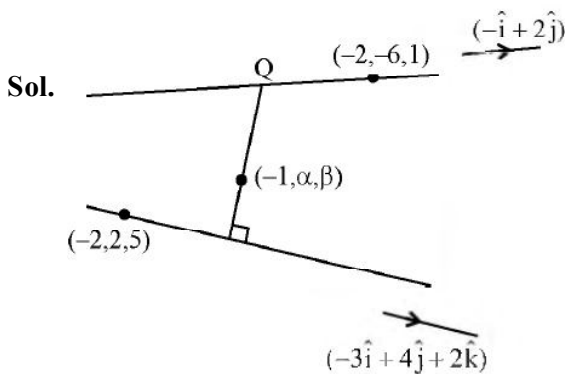
**Sol.**  $\sin^2 x - (x^2 - 2x - 2)\sin x - 3(x-1)^2 = 0$   
 $\sin^2 x - (x-1)^2 \sin x - 3(x-1)^2 = 0$



$\sin x = -3$  (reject) or  $(x-1)^2$   
 $\sin x = (x-1)^2$



24. 25



$$P(-3\lambda - 2, 4\lambda + 2, 2\lambda + 5)$$

$$Q(-\mu - 2, 2\mu - 6, 1)$$

$$\text{DRS of PQ} = \begin{vmatrix} \hat{i} & \hat{j} & \hat{k} \\ -1 & 2 & 0 \\ -3 & 4 & 2 \end{vmatrix}$$

$$= (4\hat{i} + 2\hat{j} + 2\hat{k})$$

OR

$$(2, 1, 1)$$

$$\frac{3\lambda - \mu}{2} = \frac{2\mu - 4\lambda - 8}{1} = \frac{-2\lambda - 4}{1}$$

$$\Rightarrow \mu = \lambda + 2 \text{ \& } 7\lambda = \mu - 8$$

$$\boxed{\lambda = -1} \quad \boxed{\mu = 1}$$

$$Q : (-3, -4, 1)$$

$$L_{PQ} = \frac{x+3}{2} = \frac{y+4}{1} = \frac{z-1}{1}$$

$$(-1, \alpha, \beta) \Rightarrow 1 = \frac{\alpha+4}{1} = \frac{\beta-1}{1}$$

$$\Rightarrow \alpha = -3, \beta = 2$$

$$(\alpha - \beta)^2 = 25$$

25. 76

**Sol.**  $S = 1 + \frac{x}{2\sqrt{3}} + \frac{x^3}{18} + \frac{x^3}{36\sqrt{3}} + \frac{x^4}{180} + \dots \infty$

Put  $\frac{x}{\sqrt{3}} = t$ , where  $x = \sqrt{3} - \sqrt{2}$

$$S = 1 + \frac{1}{2} + \frac{t^2}{6} + \frac{t^3}{12} + \frac{t^4}{20} + \dots$$

$$S = 1 + t\left(1 - \frac{1}{2}\right) + t^2\left(\frac{1}{2} - \frac{1}{3}\right) + t^3\left(\frac{1}{3} - \frac{1}{4}\right) + t^4\left(\frac{1}{4} - \frac{1}{5}\right)$$

$$S = \left(1 + t + \frac{t^2}{2} + \frac{t^3}{3} + \frac{t^4}{4} + \dots\right) - \left(\frac{t}{2} + \frac{t^2}{3} + \frac{t^3}{4} + \frac{t^4}{5} + \dots\right)$$

$$S = \left(t + \frac{t^2}{2} + \dots\right) - \frac{1}{t} \left(t + \frac{t^2}{2} + \frac{t^3}{3} + \dots\right) + 2$$

$$S = 2 + \left(1 - \frac{1}{t}\right) (-\log(1-t)) = \left(\frac{1}{t} - 1\right) \log(1-t) + 2$$

$$S = 2 \left( \frac{\sqrt{3}}{\sqrt{3}-\sqrt{2}} - 1 \right) \log \left( 1 - \frac{\sqrt{3}-\sqrt{2}}{\sqrt{3}} \right)$$

$$S = 2 + \left( \frac{\sqrt{2}}{\sqrt{3}-\sqrt{2}} \right) \log e \frac{\sqrt{2}}{\sqrt{3}}$$

$$S = 2 + \frac{(\sqrt{6}+2)}{3} \log e \frac{2}{3} = 2 + \left( \frac{\sqrt{3}}{2} + 1 \right) \log e \frac{2}{3}$$

$$a = 2, b = 3$$

$$11a + 18b = 11 \times 2 + 18 \times 3 = 76$$

26. 170

**Sol.**  $2x^2 + x - 2 = 0$   $\begin{matrix} \nearrow a \\ \searrow b \end{matrix}$

$$2x^2 - x - 2 = 0 \begin{matrix} \nearrow \frac{1}{a} \\ \searrow \frac{1}{b} \end{matrix}$$

$$\lim_{x \rightarrow \frac{1}{a}} \frac{\left(1 - \cos 2\left(s - \frac{1}{a}\right)\left(x - \frac{1}{b}\right)\right)}{4\left(x - \frac{1}{b}\right)} \times \frac{4\left(x - \frac{1}{b}\right)^2}{a^2\left(x - \frac{1}{a}\right)^2}$$

$$= 16 \times \frac{2}{a^2} \left(\frac{1}{a} - \frac{1}{b}\right)^2$$

$$= \frac{32}{a^2} \left(\frac{17}{4}\right) = \frac{17.8}{a^2} = \frac{17 \times 8 \times 16}{(-1 + \sqrt{117})^2}$$

$$= \frac{136.16}{18.2\sqrt{7}} \times \frac{18 + 2\sqrt{7}}{18 + 2\sqrt{7}}$$

$$= \frac{136}{256} (18 + 2\sqrt{7}) \cdot 16$$

$$= 153 + 17\sqrt{17} = \alpha + \beta\sqrt{17}$$

$$\alpha + \beta = 153 + 17 = 170$$

27. 1

**Sol.**  $f(t) = \int_0^{\pi} \frac{2x}{1 - \cos^2 t \sin^2 x} dx$  ....(1)

$$= 2 \int_0^{\pi} \frac{(\pi - x) dx}{1 - \cos^2 t \sin^2 x}$$
 ....(2)

$$2f(t) = 2 \int_0^{\pi} \frac{\pi}{1 - \cos^2 t \sin^2 x} dx$$

divide & by  $\cos^2 x$

$$f(t) = \pi \int_0^{\pi} \frac{\sec^2 x dx}{\sec^2 x - \cos^2 t \tan^2 x}$$

$$f(t) = 2\pi \int_0^{\pi/2} \frac{\sec^2 x dx}{\sec^2 x - \cos^2 t \tan^2 x}$$

$$\tan x = z$$

$$\sec^2 x dx = dz$$

$$f(t) = 2\pi \int_0^{\infty} \frac{dz}{1 + \sin^2 t \cdot z^2}$$

$$= \frac{\pi^2}{\sin t}$$

$$\text{Then } \int_0^{\pi/2} \frac{\pi^2}{f(t)} dt$$

$$= \int_0^{\pi/2} \sin t dt$$

$$= 1$$

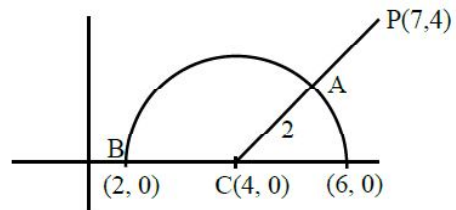
28. 1600

**Sol.**  $(x - 7)^2 + (y - 4)^2$

$$y = \sqrt{8x - x^2} - 12$$

$$y^2 = -(x - 4)^2 + 16 - 12$$

$$(x - 4)^2 + y^2 = 4$$



$$m = 9$$

$$M = 41$$

$$M^2 - m^2 = 41^2 - 9^2 = 1600$$

29. 10

**Sol.**  $y^2 = 4(x - 9)$

$$\text{slope of tangent} = \frac{-1}{2}$$

$$\text{Point of contact } P \left( 9 + \frac{1}{\left(\frac{-1}{2}\right)^2}, \frac{2 \times 1}{\frac{-1}{2}} \right)$$

$$P(13, -4)$$

$$\text{center of circle } C(7, 4)$$

$$\text{distance } CP = \sqrt{(13-7)^2 + (-4-4)^2}$$

$$= 10$$

30. 3

**Sol. Case I :**  $x \geq -5$

$$x^2 + 5x + 2x + 12 = 0$$

$$x^2 + 7x + 12 = 0$$

$$x = -3, -4$$

**Case II :**  $-7 < x < -5$

$$-x^2 - 5x + 2x + 14 - 2 = 0$$

$$-x^2 - 3x + 12 = 0$$

$$x = \frac{-3 \pm \sqrt{9+48}}{2}$$

$$= \frac{-3 \pm \sqrt{57}}{2}$$

$$x = \frac{-3 - \sqrt{57}}{2}, \frac{-3 + \sqrt{57}}{2} \text{ (rejected)}$$

**Case III :**  $x \leq -7$

$$-x^2 - 5x - 2x - 14 - 2 = 0$$

$$x^2 + 7x + 16 = 0$$

$$D = 49 - 64 < 0$$

No solutions

No. of solutions = 3

## PHYSICS

### Section - A (Single Correct Answer)

31. A

**Sol.** As  $\lambda_{\text{red}} > \lambda_{\text{yellow}} > \lambda_{\text{violet}}$

Light ray with longer wavelength bends less.

32. C

**Sol.**  $KE_{\text{max}} = hv - \phi_0 = eV$

33. A

**Sol.**  $F_c = \frac{mv^2}{r}$

$$\frac{Kq_1q_2}{r^2} = \frac{mv^2}{r}$$

$$mv^2r^2 = Kq_1q_2r$$

$$\frac{L^2}{m} = Kq_1q_2r$$

$$L \propto \sqrt{r}$$

34. C

**Sol.**  $i_g = \frac{10}{400+100} = 20 \times 10^{-3} \text{ A}$

For ammeter

Let shunt resistance = S

$$i_g R = (i - i_g) S$$

$$20 \times 10^{-3} \times 100 = 10 S$$

$$S = 20 \times 10^{-2} \Omega$$

35. A

**Sol.** Static charge is developed due to air friction. This can result in combustion. So, metallic chains is used to discharge excess charge.

36. C

**Sol.** n = number of molecule per unit volume

d = diameter of the molecule

$$\lambda = \frac{1}{\sqrt{2}\pi d^2 n} \quad (\text{By Theory})$$

37. A

**Sol.**  $\vec{P} = \cos(kt)\hat{i} - \sin(kt)\hat{j}; |\vec{P}| = 1$

$$\therefore \vec{P} = m\vec{v}$$

$$\therefore \hat{P} = \hat{v}$$

$$\Rightarrow \hat{v} = \cos(kt)\hat{i} - \sin(kt)\hat{j}$$

$$\hat{a} = \frac{-k \sin(kt)\hat{i} - k \cos(kt)\hat{j}}{k}$$

$$\Rightarrow \hat{a} = -\sin kt\hat{i} - \cos kt\hat{j}$$

$$\therefore \hat{F} = \hat{a} - \sin kt\hat{i} - \cos kt\hat{j}$$

$$\cos \theta = \frac{\hat{F} \cdot \hat{P}}{|\hat{F}| |\hat{P}|} = \frac{-\sin kt \cos t + \sin kt \cos t}{1 \times 1} = 0$$

$$\Rightarrow \theta = \frac{\pi}{2}$$

38. C

**Sol.**  $\vec{F}_1 = q\vec{E}$  (Theory)

$$\vec{F}_2 = q(\vec{V} \times \vec{B})$$

39. B

**Sol.** Given : R = 9m,  
120 revolution in 3 min

$$\omega = \frac{120 \text{ Rev.}}{3 \text{ min.}} = \frac{120 \times 2\pi \text{ rad}}{3 \times 60 \text{ sec}} = \frac{4\pi}{3} \text{ rad/s}$$

$$a_{\text{centripetal}} = \omega^2 R = \left(\frac{4\pi}{3}\right)^2 \times 9 = 16\pi^2 \text{ m/s}^2$$

40. C

**Sol.** Voltage across inductor  $V_L = IX_L$

$$31.4 = I[L\omega]$$

$$31.4 = I[L(2\pi f)]$$

$$31.4 = I[10 \times 10^{-3}(2 \times 3.14) \times 50] \Rightarrow I = 10 \text{ A}$$

41. B

**Sol.**  $\therefore [V] = [b]$

$\therefore$  Dimension of b =  $[L^3]$

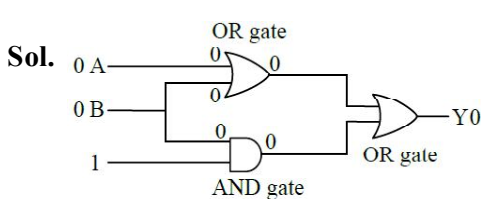
$$\&[P] = \left[\frac{a}{V^2}\right]$$

$$[a] = [PV^2] = [ML^{-1}T^{-2}][L^6]$$

$$\text{Dimension of } a = [ML^5T^{-2}]$$

$$\therefore ab^{-1} = \frac{[ML^5T^{-2}]}{[L^3]} = [ML^2T^{-2}]$$

42. B



43. C

**Sol.** P = constant  $\Rightarrow FV = \text{constant}$

$$\Rightarrow m \frac{dV}{dt} V = \text{constant}$$

$$\int_0^V V dV = (C) \int_0^t dt$$

$$\left(\frac{V^2}{2}\right) = Ct$$

$$V \propto t^{1/2}$$

$$\frac{ds}{dt} \propto t^{1/2}$$

$$\int_0^s ds = K \int_0^t t^{1/2} dt$$

$$S = K \times \frac{2}{3} t^{3/2}$$

$$S \propto t^{3/2}$$

$\therefore$  displacement is proportional to  $(t)^{3/2}$

44. B

**Sol.** Infrared is the least energetic thus having biggest wavelength ( $\lambda$ ) & gamma rays are most energetic thus having smallest wavelength ( $\lambda$ ).

45. C

**Sol.**  $P \propto T^3$

$$PT^{-3} = \text{constant}$$

$$\therefore \frac{PV}{T} = nR = \text{constant from ideal gas equation}$$

$$(P)(PV)^{-3} = \text{constant}$$

$$P^{-2} V^{-3} = \text{constant} \quad \dots(1)$$

Process equation for adiabatic process is

$$PV^y = \text{constant} \quad \dots(2)$$

Comparing equation (1) and (2)

$$\frac{C_P}{C_V} = y = \frac{3}{2}$$

46. C

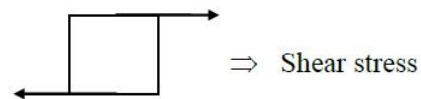
**Sol.** (A) stress =  $\frac{F_{\text{restoring}}}{A}$

If A = 1

Stress =  $F_{\text{restoring}}$

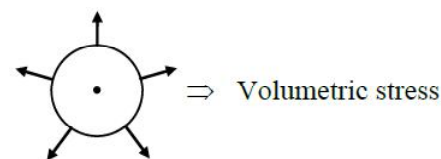
(A)-(III)

(B)



(B)-(IV)

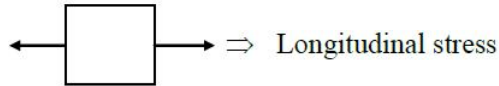
(C)



(C)-(I)



(D)



(D)-(II)

47. C

Sol. 20 VSD = 19 MSD

$$1\text{VSD} = \frac{19}{20}\text{MSD}$$

$$\text{L.C.} = 1\text{MSD} - 1\text{VSD}$$

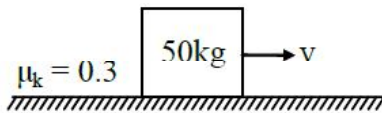
$$0.1\text{mm} = 1\text{MSD} - \frac{19}{20}\text{MSD}$$

$$0.1 = \frac{1}{20}\text{MSD}$$

$$1\text{MSD} = 2\text{mm}$$

48. B

Sol.



$$F_k = \mu_k N = 0.3 \times 50 \times 9.8 = 147\text{N}$$

49. D

Sol.  $T = \frac{2\pi r^{3/2}}{\sqrt{GM}}$

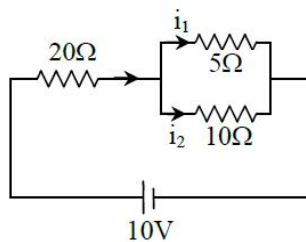
$$\frac{T_1}{T_2} = \left(\frac{r_1}{r_2}\right)^{3/2} \left(\frac{M_2}{M_1}\right)^{1/2}$$

$$\frac{6}{24} = \frac{(r_1)^{3/2}}{(4.2 \times 10^4)^{3/2}} \left(\frac{M}{M/4}\right)^{1/2}$$

$$r_1 = 1.05 \times 10^4\text{km}$$

50. B

Sol.



$$\frac{i_1}{i_2} = \frac{10}{5} = \frac{2}{1}$$

$$\frac{P_1}{P_2} = \frac{i_1^2 R_1}{i_2^2 R_2} = \left(\frac{2}{1}\right)^2 \times \frac{5}{10} = \frac{2}{1}$$

**Section - B (Numerical Value)**

51. 500

Sol.  $\mu_0 n i = B$   $n$  = number of turns per unit length

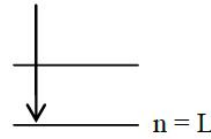
$$\mu_0 \left(\frac{m}{l}\right) i = B$$

$$m = \frac{B l}{\mu_0 i} = \frac{6.28 \times 10^{-3} \times 0.5}{12.56 \times 10^{-7} \times 5}$$

$$m = 500$$

52. 6588

Sol. Lyman Series



$$\text{Shortest, } \frac{hc}{\lambda} = -13.6 \left( \frac{1}{n_1^2} - \frac{1}{n_2^2} \right)$$

$$\lambda \uparrow E \uparrow; \frac{hc}{\lambda_0} = -13.6(1)$$

Balmer Series :

$$\text{---} \quad n = 3$$

$$\text{---} \quad n = 2$$

$$\frac{hc}{\lambda_1} = -13.6 \left( \frac{1}{2^2} - \frac{1}{3^2} \right)$$

$$\frac{hc}{\lambda_1} = -13.6 \left( \frac{1}{4} - \frac{1}{9} \right)$$

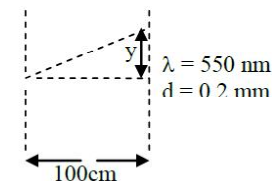
$$\frac{hc}{\lambda_1} = -13.6 \times \left( \frac{5}{36} \right)$$

$$\Rightarrow \frac{-13.6 \lambda_0}{\lambda_1} = -13.6 \times \frac{5}{36}$$

$$\lambda_1 = \frac{\lambda_0 \times 36}{5} = \frac{915 \times 36}{5} = 6588$$

53. 275

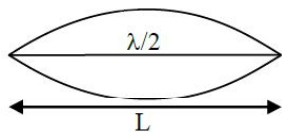
Sol.



$$y = \frac{\lambda D}{d} = \frac{550 \times 10^{-9} \times 100 \times 10^{-2}}{0.2 \times 10^{-3}} = 275$$

54. 60

Sol.



$$f_0 = 400 \text{ Hz}; v = \sqrt{\frac{T}{\mu}} = \text{constant}$$

$$\frac{\lambda}{2} = L; v = f_0 \lambda$$

$$\frac{v}{2f_0} = L \Rightarrow v = 2Lf_0$$

$$L' = \frac{v}{2f'} = \frac{2Lf_0}{2f'}$$

$$= \frac{Lf_0}{f'} = \frac{90 \times 400}{600} = 60$$

55. 2

$$\text{Sol. } \frac{\frac{1}{2}I\omega^2}{\frac{1}{2}I\omega^2 + \frac{1}{2}mv^2} = \frac{\left(\frac{1}{2}\right)\left(\frac{2}{3}mR^2\right)\omega^2}{\left(\frac{1}{2}\right)\left(\frac{2}{3}mR^2\right)\omega^2 + \frac{1}{2}m(R\omega)^2}$$

$$= \frac{\frac{2}{3}}{\frac{2}{3} + 1} = \frac{2}{5}$$

$$x = 2$$

56. 1000

$$\text{Sol. } \frac{F_1}{A_1} = \frac{F_2}{A_2}$$

$$\frac{F_1}{\pi(7)^2} = \frac{10}{\pi \times (0.7)^2}$$

$$F_1 = 1000 \text{ N}$$

57. 16

$$\text{Sol. } E_p = \frac{2KP}{r^3} = E$$

$$E_R = \frac{KP}{(2r)^3} = \frac{E}{16}$$

$$x = 16$$

58. 16

$$\text{Sol. } H_{\max} = \frac{u^2 \sin^2 \theta}{2g}$$

$$\frac{H_{1\max}}{H_{2\max}} = \frac{u_1^2}{u_2^2}$$

$$\frac{64}{H_{2\max}} = \frac{u^2}{(u/2)^2}$$

$$H_{2\max} = 16 \text{ m}$$

59. 5

$$\text{Sol. } \frac{20\Omega}{5} \Rightarrow 10 \text{ equal part}$$

Each part has resistance =  $2\Omega$

2 parts are connected in parallel so,  $R = 1\Omega$

Now, there will be 5 parts each of resistance  $1\Omega$ , they are connected in series.

$$R_{\text{eq}} = 5R, R_{\text{eq}} = 5\Omega$$

60. 4

$$\text{Sol. } I = 3t + 8$$

$$\varepsilon = 12 \text{ mV}$$

$$|\varepsilon| = L \left| \frac{dI}{dt} \right|$$

$$12 = L \times 3$$

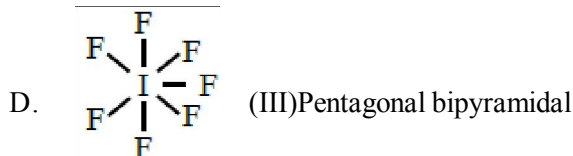
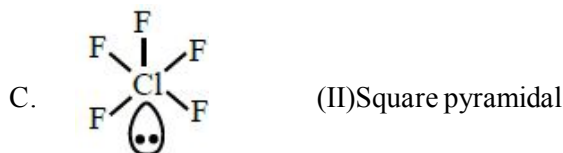
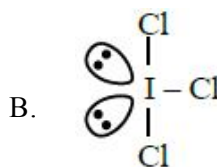
$$L = 4 \text{ mH}$$

## CHEMISTRY

### Section - A (Single Correct Answer)

61. (C)

Sol. A. I - Cl (IV) linear



62. (A)

**Sol.**  $\text{Fe}^{2+}$  ions undergoes hydrolysis, therefore while preparing aqueous solution of ferrous sulphate and ammonium sulphate in water dilute sulphuric acid is added to prevent hydrolysis of ferrous sulphate.

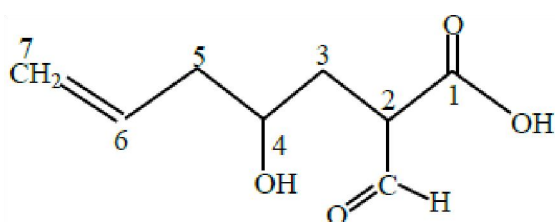
63. (C)

**Sol.:**



64. (C)

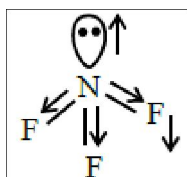
**Sol.:**



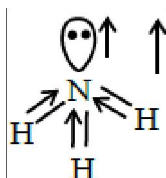
2-formyl-4-hydroxyhept-6-enoic acid

65. (A)

**Sol.**



Resultant dipole moment =  $0.80 \times 10^{-30}$  Cm



Resultant dipole moment =  $4.90 \times 10^{-30}$  cm

66. (A)

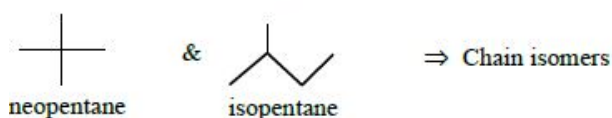
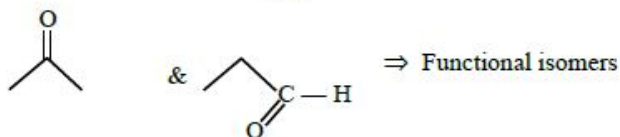
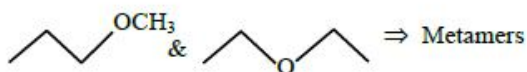
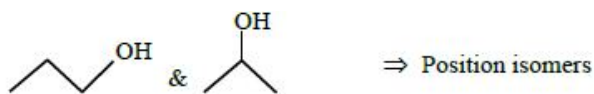
**Sol.**  $\text{BaCl}_2$ ,  $\text{NaCl}$  are soluble but on adding  $\text{HCl}(\text{g})$  to  $\text{BaCl}_2$ ,  $\text{NaCl}$  solutions, Sodium or Barium chlorides may precipitate out, as a consequence of the law of mass action.

67. (B)

**Sol.** Tetrahedral complex does not show geometrical isomerism.

68. (D)

**Sol.:**



69. (D)

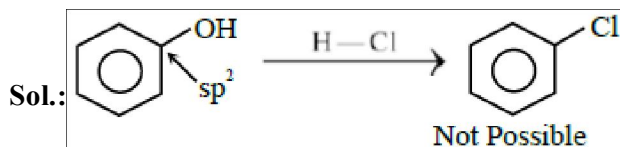
**Sol.**  $W = ZIt$

$W = ZQ$

$$Q = \frac{W}{Z}$$

$W = ZQ = (\text{electrochemical equivalent})$

70. (B)



71. (A)

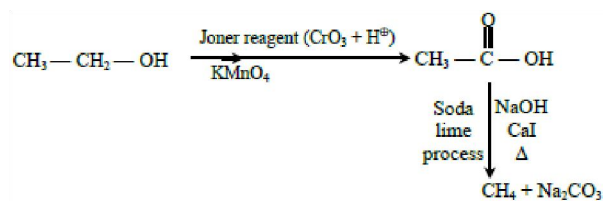
**Sol.**  $r_{\text{Na}} > r_{\text{Na}^+}$

So, Statement (I) is correct but size of anions are greater than size of neutral atoms.

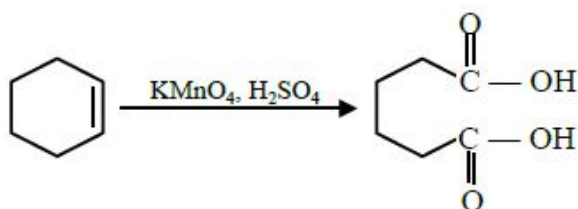
So statement (II) is incorrect.

72. (A)

**Sol.**



73. (D)



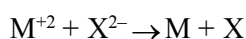
74. (C)

Sol.  $M | M^{+2} || X / X^{2-}$

$$E_{\text{cell}}^0 = E_{M/M^{+2}}^0 + E_{X^{2-}/X}^0$$

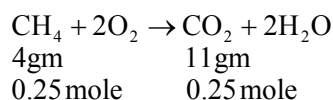
$$= -0.46 + 0.34 = -0.12\text{V}$$

As  $E_{\text{cell}}^0$  is negative so anode becomes cathode and cathode become anode. Spontaneous reaction will be



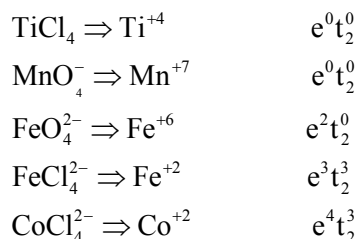
75. (B)

Sol.  $C_n H_{2n+2} + \frac{3n+1}{2} O_2 \rightarrow nCO_2 + (n+1)H_2O$



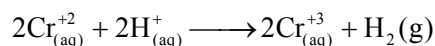
0.25 mole  $CH_4$  gives 0.25 mole (or 11 gm)  $CO_2$

76. (A)



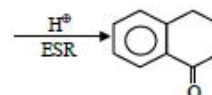
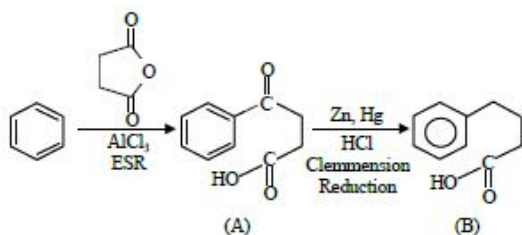
77. (C)

Sol. The ions  $Ti^{+2}$ ,  $V^{+2}$ ,  $Cr^{+2}$  are strong reducing agents and will liberate hydrogen from a dilute acid, eg.



78. (B)

Sol.



79. (A)

Sol. A. size order  $T \ell > In > Al > Ga > B$

B. Electronegativity order  $B > Al < Ga < In < T \ell$

D. B, Al are more stable in +3 oxidation state

So, only C, E statements are correct.

80. (A)

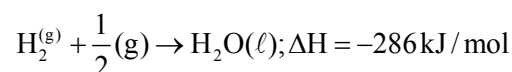
Sol. Coagulation of egg give primary structure of protein, which is known as denaturation of protein

### Section - B (Numerical Value Type)

81. (6535)

Sol.  $6C(\text{graphite}) + 3H_2(g) \rightarrow C_6H_6(\ell); \Delta H = 48.5 \text{ kJ/mol}$

$C(\text{graphite}) + O_2(g) \rightarrow CO_2(g); \Delta H = -393.5 \text{ kJ/mol}$



equation  $-(1) \times 1 + (2) \times 6 + (3) \times 3$

$-48.5 - 6 \times 393.5 - 3 \times 286$

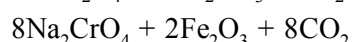
$= -3267.5 \text{ kJ for 1 mol}$

$= -6535 \text{ kJ for 2 mol}$

Ans. 6535 kJ

82. (6)

Sol.  $4FeCr_2O_4 + 8Na_2CO_3 + 7O_2 \rightarrow$



A                      B

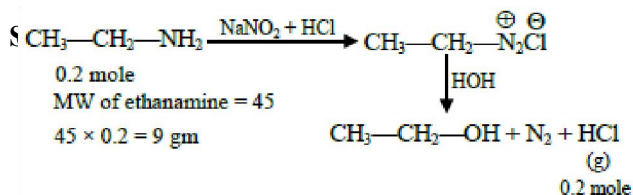
Spin only magnetic moment

For  $Na_2CrO_4$                        $\mu_B = 0$

For  $Fe_2O_3$                        $\mu_B = 5.9$

sum = 5.9

83. (9)



84. (6)

Sol.  $n = 4$

$l$	$m_l$
0	0
1	-1, 0, +1
2	-2, -1, 0, +1, +2, +3

So number of orbital associated with  $n = 4$ ,  $|m_l| = 1$  are 6

Now each orbital contain one  $e^-$  with  $m_s = -\frac{1}{2}$

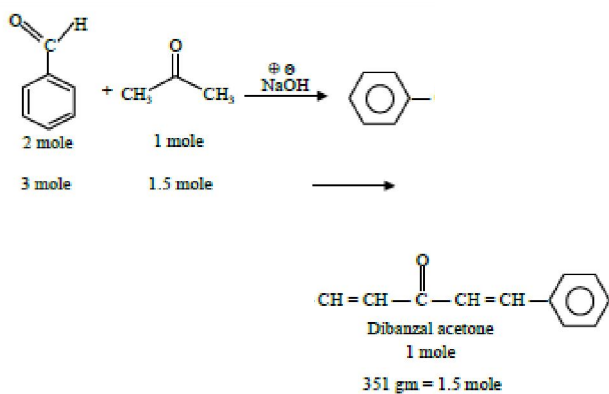
85. (50)

Sol.:  $R_f$  of A =  $\frac{5}{12.5}$       $R_f$  of C =  $\frac{10}{12.5}$

$$\text{Ratio} = \frac{R_{f(A)}}{R_{f(C)}} = \frac{1}{2} = 0.5 \text{ or } 50 \times 10^{-2}$$

86. (318)

Sol. Claisen Schmidt reaction



mw of benzaldehyde = 106

$106 \times 3 = 318$  gm. Benzaldehyde is required to give 1.5 mole (or 351 gm) product.

87. (315)

Sol.  $2A(g) + B(g) \rightarrow C(g)$

$r_1$  1.5 atm     0.7 atm  
 $r_2$  0.5 atm     0.2 atm     0.5 atm

$$\therefore r = K [P_A]^2 [P_B]$$

$$r_1 = K [1.5]^2 [0.7]$$

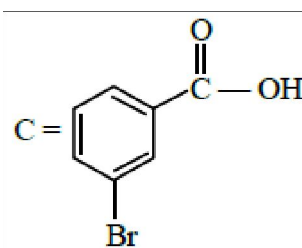
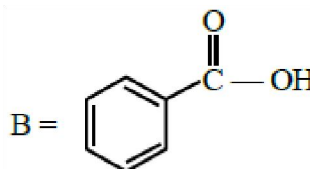
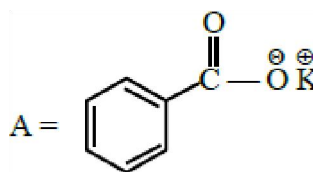
$$r_2 = K [0.5]^2 [0.2]$$

$$\frac{r_1}{r_2} = 9 \times \frac{7}{2} = 31.5 = 315 \times 10^{-1}$$

Ans. 315

88. (4)

Sol.



$\pi$  bonds = 4

89. (19)

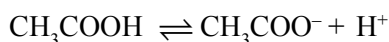
Sol.: Mass of  $\text{CH}_3\text{COOH} = V \times d$

$$= 5 \text{ ml} \times 1.2 \text{ g/ml}$$

$$= 6 \text{ gm}$$

$$n\text{CH}_3\text{COOH} = \frac{6}{60} = 0.1 \text{ mol}$$

$$m\text{CH}_3\text{COOH} \approx M_{\text{CH}_3\text{COOH}} = \frac{0.1}{1} = 0.1 \text{ M}$$



C

C- $\alpha$                        $\text{C}\alpha$                        $\text{C}\alpha$

$$K_a = \frac{C\alpha^2}{1-\alpha}$$

$$1-\alpha \approx 1 \Rightarrow K_a = C\alpha^2$$

$$\alpha = \sqrt{\frac{K_a}{C}} = \sqrt{\frac{6.25 \times 10^{-5}}{0.1}} = 25 \times 10^{-3}$$

$$V.f.(i) = 1 + \alpha(n-1) = 1 + \alpha(2-1) = 1 + \alpha = 1 + 25 \times 10^{-3} = 1.025$$

$$\Delta T_f = iK_f m$$

$$= (1.025) (1.86) (0.1)$$

$$= 0.19$$

$$= 19 \times 10^{-2}$$

90. (6)

**Sol.**  $H_2$ ,  $CO_2$ ,  $BF_3$ ,  $CH_4$ ,  $SiF_4$ ,  $BeF_2$

are symm. molecule so dipole moment is zero

