

**MATHEMATICS****Section - A (Single Correct Answer)**

1. D

**Sol.** 
$$\begin{vmatrix} 1 & 1 & 1 \\ 1 & 2 & \lambda^2 \\ 1 & 3 & \lambda \end{vmatrix} = 0$$

$$\Rightarrow 2\lambda^2 - \lambda - 1 = 0$$

$$\lambda = 1, -\frac{1}{2}$$

$$\begin{vmatrix} 1 & 1 & 5 \\ 2 & \lambda^2 & 9 \\ 3 & \lambda & \mu \end{vmatrix} = 0 \Rightarrow \mu = 13$$

Infinite solution  $\lambda = 1$  &  $\mu = 13$ For unique solution  $\lambda \neq 1$ For solution  $\lambda = 1$  &  $\mu \neq 13$ If  $\lambda \neq 1$  &  $\mu \neq 13$ Considering the case when  $\lambda = -\frac{1}{2}$  and  $\mu \neq 13$ 

this will generate no solution case

2. B

**Sol.**  $3\sin\alpha\cos\beta + 3\sin\beta\cos\alpha$

$$= 2\sin\alpha\cos\beta - 2\sin\beta\cos\alpha$$

$$5\sin\beta\cos\alpha = -\sin\alpha\cos\beta$$

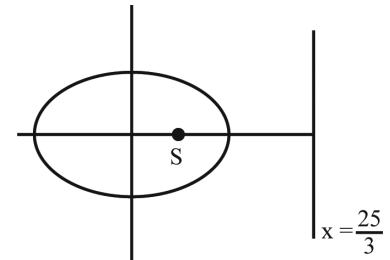
$$\tan\beta = -\frac{1}{5}\tan\alpha$$

$$\tan\alpha = -5\tan\beta$$

3. D

**Sol.**  $A = (10, 0)$

$$B\left(0, \frac{50}{7}\right) \quad P = (3, 5)$$



$$ae = 3$$

$$\frac{a}{e} = \frac{25}{3}$$

$$a = 5$$

$$b = 4$$

$$\text{Length of LR} = \frac{2b^2}{a} = \frac{32}{5}$$

4. A

**Sol.**  $|\vec{b}|^2 = 6$ ;  $|\vec{a}| |\vec{b}| \cos\theta = 3\sqrt{2}$

$$|\vec{a}|^2 |\vec{b}|^2 \cos^2 \theta = 18$$

$$|\vec{a}|^2 = 6$$

$$\text{Also } 1 + \alpha^2 + \beta^2 = 6$$

$$\alpha^2 + \beta^2 = 5$$

to find

$$(\alpha^2 + \beta^2) |\vec{a}|^2 |\vec{b}|^2 \sin^2 \theta$$

$$= (5)(6)(6)\left(\frac{1}{2}\right)$$

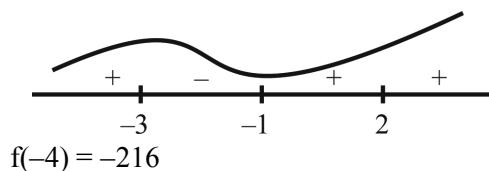
$$= 90$$

5. C

**Sol.**  $f'(x) = (x+3)^2 \cdot 3(x-2)^2 + (x-2)^3 \cdot 2(x+3)$

$$= 5(x+3)(x-2)^2(x+1)$$

$$f'(x) = 0, x = -3, -1, 2$$



$$f(-4) = -216$$

$$f(-3) = 0, f(4) = 49 \times 8 = 392$$

$$M = 392, m = -216$$

$$M - m = 392 + 216 = 608$$

Ans = '3'

6. C

$$\text{Sol. } 1^{\text{st}} \text{ GP} \Rightarrow t_1 = a, t_3 = b = ar^2 \Rightarrow r^2 = \frac{b}{a}$$

$$t_{11} = ar^{10} = a(r^2)^5 = a \left(\frac{b}{a}\right)^5$$

$$2^{\text{nd}} \text{ G.P.} \Rightarrow T_1 = a, T_5 = ar^4 = b$$

$$\Rightarrow r^4 = \left(\frac{b}{a}\right) \Rightarrow r = \left(\frac{b}{a}\right)^{1/4}$$

$$T_p = ar^{p-1} = a \left(\frac{b}{a}\right)^{\frac{p-1}{4}}$$

$$t_{11} = T_p \Rightarrow a \left(\frac{b}{a}\right)^5 = a \left(\frac{b}{a}\right)^{\frac{p-1}{4}}$$

$$\Rightarrow 5 = \frac{p-1}{4} \Rightarrow p = 21$$

7. A

**Sol.** Cocus of point P(x, y) whose distance from  
Gives

$X + 2y + 7 = 0$  &  $2x - y + 8 = 0$  are equal is

$$\frac{x+2y+7}{\sqrt{5}} = \pm \frac{2x-y+8}{\sqrt{5}}$$

$$(x+2y+7)^2 - (2x-y+8)^2 = 0$$

Combined equation of lines

$$(x-3y+1)(3x+y+15)=0$$

$$3x^2 - 3y^2 - 8xy + 18x - 44y + 15 = 0$$

$$x^2 - y^2 - \frac{8}{3}xy + 6x - \frac{44}{3}y + 5 = 0$$

$$x^2 - y^2 + 2hxy + 2gx + 2fy + c = 0$$

$$h = \frac{4}{3}, g = 3, f = -\frac{22}{3}, c = 5$$

$$g+c+h-f = 3+5-\frac{4}{3}+\frac{22}{3} = 8+6 = 14$$

8. B

$$\text{Sol. } |\vec{b}| = 1 \text{ & } |\vec{b} \times \vec{a}| = 2$$

$$(\vec{b} \times \vec{a}) \cdot \vec{b} = \vec{b} \cdot (\vec{b} \times \vec{a}) = 0$$

$$|(\vec{b} \times \vec{a}) - \vec{b}|^2 = |\vec{b} \times \vec{a}|^2 + |\vec{b}|^2$$

$$= 4 + 1 = 5$$

9. B

$$\text{Sol. } y = f(x) \Rightarrow \frac{dy}{dx} = f'(x)$$

$$\left. \frac{dy}{dx} \right|_{(1, f(1))} = f'(1) = \tan \frac{\pi}{6} = \frac{1}{\sqrt{3}} \Rightarrow f'(1) = \frac{1}{\sqrt{3}}$$

$$\left. \frac{dy}{dx} \right|_{(3, f(3))} = f'(3) = \tan \frac{\pi}{4} = 1 \Rightarrow f'(3) = 1$$

$$27 \int_1^3 ((f'(t))^2 + 1) f''(t) dt = \alpha + \beta \sqrt{3}$$

$$I = \int_1^3 ((f'(t))^2 + 1) f''(t) dt$$

$$f'(1) = z \Rightarrow f''(t) dt = dz$$

$$z = f'(3) = 1$$

$$z = f'(1) = \frac{1}{\sqrt{3}}$$

$$I = \int_{1/\sqrt{3}}^1 (z^2 + 1) dz = \left( \frac{z^3}{3} + z \right)_{1/\sqrt{3}}^1$$

$$= \left( \frac{1}{3} + 1 \right) - \left( \frac{1}{3} \cdot \frac{1}{3\sqrt{3}} + \frac{1}{\sqrt{3}} \right)$$

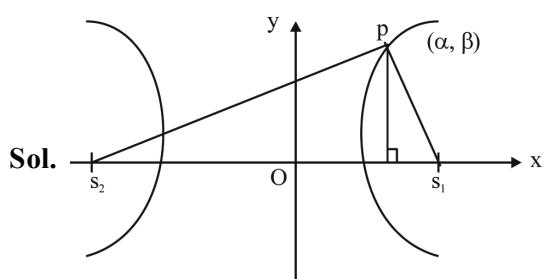
$$= \frac{4}{3} - \frac{10}{9\sqrt{3}} = \frac{4}{3} - \frac{10}{27}\sqrt{3}$$

$$\alpha + \beta \sqrt{3} = 27 \left( \frac{4}{3} - \frac{10}{27}\sqrt{3} \right) = 36 - 10\sqrt{3}$$

$$\alpha = 36, \beta = -10$$

$$\alpha + \beta = 36 - 10 = 26$$

10. C



**Sol.**

$$\frac{x^2}{9} - \frac{y^2}{4} = 1$$

$$a^2 = 9, b^2 = 4$$

$$b^2 = a^2(e^2 - 1) \Rightarrow e^2 = 1 + \frac{b^2}{a^2}$$

$$e^2 = 1 + \frac{4}{9} = \frac{13}{9}$$

$$e = \frac{\sqrt{13}}{3} \Rightarrow s_1 s_2 = 2ae = 2 \times 3 \times \sqrt{\frac{13}{3}} = 2\sqrt{13}$$

$$\text{Area of } \Delta P S_1 S_2 = \frac{1}{2} \times \beta \times s_1 s_2 = 2\sqrt{13}$$

$$\Rightarrow \frac{1}{2} \times \beta \times (2\sqrt{13}) = 2\sqrt{13} \Rightarrow \beta = 2$$

$$\frac{\alpha^2}{9} - \frac{\beta^2}{4} = 1 \Rightarrow \frac{\alpha^2}{9} - 1 = 1 \Rightarrow \alpha^2 = 18 \Rightarrow \alpha = 3\sqrt{2}$$

$$\text{Distance of } P \text{ from origin} = \sqrt{\alpha^2 + \beta^2}$$

$$= \sqrt{18+4} = \sqrt{22}$$

11. C

**Sol.**  $E_1 : A$  is selected      

A
3W
7R

B
3W
2R

$E_2 : B$  is selected

E : white ball is drawn

$$P(E_1/E) =$$

$$\frac{P(E) \cdot P(E/E_1)}{P(E_1) \cdot P(E/E_1) + P(E_2) \cdot P(E/E_2)} = \frac{\frac{1}{2} \times \frac{3}{10}}{\frac{1}{2} \times \frac{3}{10} \times \frac{1}{2} \times \frac{3}{5}}$$

$$= \frac{3}{3+6} = \frac{1}{3}$$

12. D

**Sol.**  $f(x) = ae^{2x} + be^x + cx$

$$f(0) = -1$$

$$a + b = -1$$

$$f'(x) = 2ae^{2x} + be^x + c$$

$$f'(\ln 2) = 21$$

$$8a + 2b + c = 21$$

$$\int_0^{\ln 4} (ae^{2x} + be^x) dx = \frac{39}{2}$$

$$\left[ \frac{ae^{2x}}{2} + be^x \right]_0^{\ln 4} = \frac{39}{2} \Rightarrow 8a + 4b - \frac{a}{2} - b = \frac{39}{2}$$

$$15a + 6b = 39$$

$$15a - 6a - 6 = 39$$

$$9a = 45 \Rightarrow a = 5$$

$$b = -6$$

$$c = 21 - 40 + 12 = -7$$

$$a + b + c = 8$$

$$|a + b + c| = 8$$

13. A

**Sol.**  $L_1 \perp L_2 \quad L_3 \perp L_1, L_2$

$$3 - 1 + 2P = 0$$

$$P = -1$$

$$\begin{vmatrix} \hat{i} & \hat{j} & \hat{k} \\ 1 & -1 & 2 \\ 3 & 1 & -1 \end{vmatrix} = -\hat{i} + 7\hat{j} + 4\hat{k}$$

$\therefore (-\delta, 7\delta, 4\delta)$  will lie on  $L_3$

For  $\delta = 1$  the point will be  $(-1, 7, 4)$

14. D

**Sol.** f is continuous  $f'(x) = 2x + 3, k < 1$

$$\therefore 4 + a = b + 2 \quad b, \quad x > 1$$

$$a = b = 2 \quad f \text{ is differentiable}$$

$$\therefore b = 5$$

$$\therefore a = 3$$

$$\int_{-2}^1 (x^2 + 3x^3) dx + \int_1^2 (5x + 2) dx$$

$$= \left[ \frac{x^3}{3} + \frac{3x^2}{2} + 3x \right]_{-2}^1 + \left[ \frac{5x^2}{2} + 2x \right]_1^2$$

$$= \left( \frac{1}{3} + \frac{3}{2} + 3 \right) - \left( \frac{-8}{3} + 6 - 6 \right) + \left( 10 + 4 - \frac{5}{2} - 2 \right)$$

$$= 6 + \frac{3}{2} + 12 - \frac{5}{2} = 17$$

15. A

**Sol.**  $f\left(\frac{x}{y}\right) = \frac{f(x)}{f(y)}$   $f'(1) = 2024$

$$f(1) = 1$$

Partially differentiating w.r.t. x

$$f'\left(\frac{x}{y}\right) \cdot \frac{1}{y} = \frac{1}{f(y)} f'(x)$$

$y \rightarrow x$

$$f'(1) \cdot \frac{1}{x} = \frac{f'(x)}{f(x)}$$

$$2024f(x) = xf'(x) \Rightarrow xf'(x) - 2024f(x) = 0$$

16. B

**Sol.**  $z^{1985} + z^{100} + 1 = 0$  &  $z^3 + 2z^2 + 2z + 1 = 0$

$$(z+1)(z^2-z+1) + 2z(z+1) = 0$$

$$(z+1)(z^2+z+1) = 0$$

$$\Rightarrow z = -1, z = w, w^2$$

Now putting  $z = -1$  not satisfy

Now put  $z = w$

$$\Rightarrow w^{1985} + w^{100} + 1$$

$$\Rightarrow w^2 + w + 1 = 0$$

Also,  $z = w^2$

$$\Rightarrow w^{3970} + w^{200} + 1$$

$$\Rightarrow w + w^2 + 1 = 0$$

Two common root

17. Bonus

**Sol.**  $2-p, p, 2-\alpha, \alpha$

Binomial coefficients are

$nC_r, nC_{r+1}, nC_{r+2}, nC_{r+3}$  respectively

$$\Rightarrow nC_r + nC_{r+1} = 2$$

$$\Rightarrow n+1C_{r+1} = 2 \dots\dots(1)$$

Also,  $nC_{r+2} + nC_{r+3} = 2$

$$\Rightarrow n+1C_{r+3} = 2 \dots\dots(2)$$

From (1) and (2)

$$n+1C_{r+1} = n+1C_{r+3}$$

$$\Rightarrow 2r+4 = n+1$$

$$n = 2r+3$$

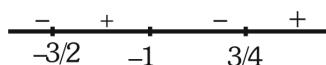
$$2r+4C_{r+1} = 2$$

Data Inconsistent

18. B

**Sol.**  $\frac{2x+3}{4x^2+x-3} > 0$  and  $-1 \leq \frac{2x-1}{x+2} \leq 1$

$$\frac{2x+3}{(4x-3)(x+1)} > 0 \quad \frac{3x+1}{x+2} \geq 0 \quad \text{and} \quad \frac{x-3}{x+2} \leq 0$$



$$(-\infty, -2) \cup \left[ \frac{-1}{3}, \infty \right) \dots\dots(1)$$

$$(-2, 3] \dots\dots(2)$$

$$\left[ \frac{-1}{3}, 3 \right] \dots\dots(3) \quad (1) \cap (2) \cap (3)$$

$$\left( \frac{3}{4}, 3 \right]$$

$$\alpha = \frac{3}{4} \quad \beta = 3$$

$$5\beta - 4\alpha = 15 - 3 = 12$$

19. D

**Sol.**  $f(x) = \frac{x}{(1+x^4)^{1/4}}$

$$f \circ f(x) = \frac{f(x)}{(1+f(x)^4)^{1/4}} = \frac{\frac{x}{(1+x^4)^{1/4}}}{\left(1+\frac{x^4}{1+x^4}\right)^{1/4}} = \frac{x}{(1+2x^4)^{1/4}}$$

$$f(f(f(f)x)) = \frac{x}{(1+4x^4)^{1/4}}$$

$$18 \int_0^{\sqrt[4]{5}} \frac{x^3}{(1+4x^4)^{1/4}} dx$$

$$\text{Let } 1+4x^4 = t^4$$

$$16x^3 dx = 4t^3 dt$$

$$\frac{18}{4} \int_1^3 \frac{t^3 dt}{t}$$

$$= \frac{9}{2} \left( \frac{t^3}{3} \right)_1^3$$

$$= \frac{3}{2}[26] = 39$$

20. B

**Sol.**  $x = \sin \theta = y \sin \left( \theta + \frac{2\pi}{3} \right) = z \sin \left( \theta + \frac{4\pi}{3} \right) \neq 0$

$$\Rightarrow x, y, z \neq 0$$

Also,

$$\sin \theta + \sin\left(\theta + \frac{2\pi}{3}\right) + \sin\left(\theta + \frac{4\pi}{3}\right) = 0 \quad \forall \theta \in \mathbb{R}$$

$$\Rightarrow \frac{1}{x} + \frac{1}{y} + \frac{1}{z} = 0$$

$$\Rightarrow xy + yz + zx = 0$$

(i) Trace (R) = x + y + z

If x + y + z = 0 and xy + yz + zx = 0

$$\Rightarrow x = y = z = 0$$

Statement (i) is False

(ii) Adj(Adj(R)) = |R| R

Trace (Adj(Adj(R)))

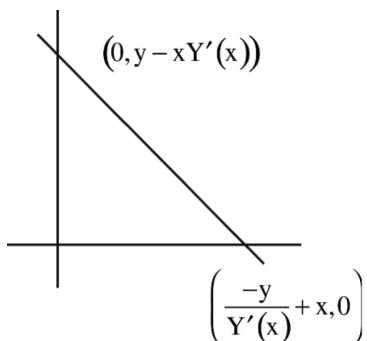
$$= xyz(x + y + z) \neq 0$$

Statement (ii) is also False

### Section - B (Numerical Value Type)

21. 20

**Sol.**  $A = \frac{1}{2} \left( \frac{-y}{Y'(x)} + x \right) (y - xY/x) = \frac{-y^2}{2Y'(x)} + 1$



$$\Rightarrow (-y + xY'(x))(y - xY'(x)) = -y^2 + 2Y'(x)$$

$$-y^2 + xyY'(x) + xyY'(x) - x^2 [Y'(x)]^2$$

$$= -y^2 + 2Y'(x)$$

$$2xy - x^2 Y'(x) = 2$$

$$\frac{dy}{dx} = \frac{2xy - 2}{x^2}$$

$$\text{I.F.} = e^{-2\ln x} = \frac{1}{x^2}$$

$$y \cdot \frac{1}{x^2} = \frac{2}{3}x^{-3} + c$$

$$\text{Put } x = 1, y = 1$$

$$1 = \frac{2}{3} + c \Rightarrow c = \frac{1}{3}$$

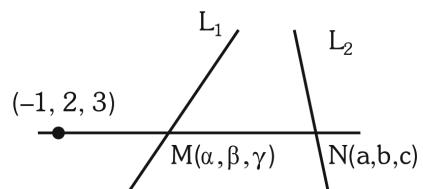
$$Y = \frac{2}{3} \cdot \frac{1}{x} + \frac{1}{3} X^2$$

$$\Rightarrow 12Y(2) = \frac{5}{3} \times 12 = 20$$

22. 196

**Sol.**  $M(3\lambda+1, 2\lambda+2, -2\lambda-1) \therefore \alpha + \beta + \gamma = 3\lambda + 2$

$$N(-3\mu-2, -2\mu+2, 4\mu+1) \therefore a+b+c = -\mu+1$$



$$\frac{3\lambda+2}{-3\mu-1} = \frac{2\lambda}{-2\mu} = \frac{-2\lambda-4}{4\mu-2}$$

$$3\lambda\mu + 2\mu = 3\lambda\mu + \lambda$$

$$2\mu = \lambda$$

$$2\lambda\mu - \lambda = \lambda\mu + 2\mu$$

$$\lambda\mu = \lambda + 2\mu$$

$$\Rightarrow \lambda\mu = 2\lambda$$

$$\Rightarrow \mu = 2 \quad (\lambda \neq 0)$$

$$\therefore \lambda = 4$$

$$\alpha + \beta + \gamma = 14$$

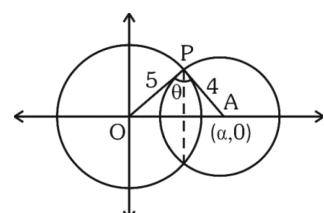
$$a + b + c = -1$$

$$\frac{(\alpha + \beta + \gamma)^2}{(a + b + c)^2} = 196$$

23. 1575

**Sol.**  $C_1 : x^2 + y^2 = 25, C_2 : (x - \alpha)^2 + y^2 = 16$

$$5 < \alpha < 9$$



$$\theta = \sin^{-1} \left( \frac{\sqrt{63}}{8} \right)$$

$$\sin \theta = \frac{\sqrt{63}}{8}$$

$$\text{Area of } \Delta OAP = \frac{1}{2} \times \alpha \left( \frac{\beta}{2} \right) = \frac{1}{2} \times 5 \times 4 \sin \theta$$

$$\Rightarrow \alpha\beta = 40 \times \frac{\sqrt{63}}{8}$$

$$\alpha\beta = 5 \times \sqrt{63}$$

$$(\alpha\beta)^2 = 25 \times 63 = 1575$$

24. 10

$$\text{Sol. } \alpha = \sum_{k=0}^n \frac{{}^n C_k \cdot {}^n C_k}{k+1} \cdot \frac{n+1}{n+1}$$

$$= \frac{1}{n+1} \sum_{k=0}^n {}^{n+1} C_{k+1} \cdot {}^n C_{n-k}$$

$$\alpha = \frac{1}{n+1} \cdot {}^{2n+1} C_{n+1}$$

$$\beta = \sum_{k=0}^{n-1} {}^n C_k \cdot \frac{{}^n C_{k+1}}{k+2} \frac{n+1}{n+1}$$

$$= \frac{1}{n+1} \sum_{k=0}^{n-1} {}^n C_{n-k} \cdot {}^{n+1} C_{k+2}$$

$$= \frac{1}{n+1} \cdot {}^{2n+1} C_{n+2}$$

$$\frac{\beta}{\alpha} = \frac{{}^{2n+1} C_{n+2}}{{}^{2n+1} C_{n+1}} = \frac{2n+1-(n+2)+1}{n+2}$$

$$\frac{\beta}{\alpha} = \frac{n}{n+2} = \frac{5}{6}$$

$$n = 10$$

25. 9

**Sol.**  $S_n = 3 \cdot 7 + 11 + \dots + n$  terms

$$= \frac{n}{2} (6 + (n-1)4) = 3n + 2n^2 - 2n$$

$$= 2n^2 + n$$

$$\sum_{k=1}^n S_k = 2 \sum_{k=1}^n K^2 + \sum_{k=1}^n K$$

$$= 2 \cdot \frac{n(n+1)(2n+1)}{6} + \frac{n(n+1)}{2}$$

$$= n(n+1) \left[ \frac{2n+1}{3} + \frac{1}{2} \right]$$

$$= \frac{n(n+1)(4n+5)}{6}$$

$$\Rightarrow 40 < \frac{6}{n(n+1)} \sum_{k=1}^n S_k < 42$$

$$40 < 4n+5 < 42$$

$$35 < 4n < 37$$

$$n = 9$$

26. 11376

**Sol.** If 4 questions from each section are selected  
Remaining 3 questions can be selected either in  
(1, 1, 1) or (3, 0, 0) or (2, 1, 0)

$$\begin{aligned} \therefore \text{Total ways} &= {}^8 C_5 \cdot {}^6 C_5 + {}^8 C_6 \cdot {}^6 C_5 \cdot {}^6 C_4 \times 2 \\ &+ {}^8 C_6 \cdot {}^6 C_4 \times 2 + {}^8 C_4 \cdot {}^6 C_6 \cdot {}^6 C_5 \times 2 + {}^8 C_7 \cdot {}^6 C_4 \\ &\cdot {}^6 C_4 \\ &= 56 \cdot 6 \cdot 6 + 28 \cdot 6 \cdot 15 \cdot 2 + 56 \cdot 15 \cdot 2 + 70 \cdot 6 \\ &\cdot 2 + 15 \cdot 15 \\ &= 2016 + 5040 + 1680 + 840 + 1800 = 11376 \end{aligned}$$

27. 960

**Sol.** Total number of relation both symmetric and  
reflexive  $= 2^{\frac{n^2-n}{2}}$

$$\text{Total number of symmetric relation} = 2^{\frac{n^2+n}{2}}$$

$\Rightarrow$  Then number of symmetric relation which  
are not reflexive

$$\Rightarrow 2^{\frac{n(n+1)}{2}} - 2^{\frac{n(n-1)}{2}}$$

$$\Rightarrow 2^{10} - 2^6$$

$$\Rightarrow 1024 - 64$$

$$= 960$$

28. 1

**Sol.**  $x = 0$  and  $x^2 + 3|x| + 5|x-1| + 6|x-2| = 0$

Here all terms are +ve except at  $x = 0$

So there is no value of  $x$

Satisfies this equation

Only solution  $x = 0$

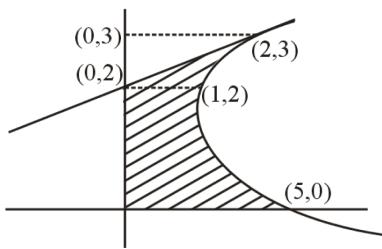
No of solution 1.

29. 5

**Sol.** Solving the equations

$$(y-2)^2 = x-1 \text{ and } x-2y+4=0$$

$$X = 2(y-2)$$



$$\frac{x^2}{4} = x - 1$$

$$x^2 - 4x + 4 = 0$$

$$(x - 2)^2 = 0$$

$$x = 2$$

Enclosed area (w.r.t. y-axis) =  $\int_0^3 x \, dy - \text{Area of } \Delta.$

$$= \int_0^3 ((y-2)^2 + 1) \, dy - \frac{1}{2} \times 1 \times 2$$

$$= \int_0^3 (y^2 - 4y + 5) \, dy - 1$$

$$= \left[ \frac{y^3}{3} - 2y^2 + 5y \right]_0^3 - 1$$

$$= 9 - 18 + 15 - 1 = 5$$

30. 29

$x_i$	$f_i$	$f_i x_i$	$f_i x_i^2$
0	3	0	0
1	2	2	2
5	3	15	75
6	2	12	72
10	6	60	600
12	3	36	432
17	3	51	867
	$\sum f_i = 22$		$\sum f_i x_i^2 = 2048$

$$\therefore \sum f_i x_i = 176$$

$$\text{So } \bar{x} = \frac{\sum f_i x_i}{\sum f_i} = \frac{176}{22} = 8$$

$$\text{for } \sigma^2 = \frac{1}{N} \sum f_i x_i^2 - (\bar{x})^2$$

$$= \frac{1}{22} \times 2048 - (8)^2$$

$$= 93.090964$$

$$= 29.0909$$

## PHYSICS

### Section - A (Single Correct Answer)

31. D

$$\text{Sol. } 50V + S = 49S + S$$

$$S = 50(S - V)$$

$$.5 = 50(S - V)$$

$$S - V = \frac{0.5}{50} = \frac{1}{100} = 0.01 \text{ mm}$$

32. B

**Sol.** Work done again frictional force

$$= \mu N \times 10$$

$$= 0.1 \times 5 \times 10 = 5J$$

33. D

**Sol.** K.E. =  $hf - \phi$

$$\tan \theta = h$$

34. D

35. C

**Sol.**  $(X) \rightarrow (Y) + (Z) + (P)$

$$M \quad M/3 \quad M/3 \quad M/3$$

$$\Delta Mc^2 = \frac{1}{2} \frac{M}{3} V^2 + \frac{1}{2} \frac{M}{3} V^2 + \frac{1}{2} \frac{M}{3} V^2$$

$$V = c \sqrt{\frac{2\Delta M}{M}}$$

36. A & C

**Sol.** Steeper curve (B) is adiabatic

Adiabatic  $\Rightarrow PV^\gamma = \text{const.}$

$$\text{Or } P \left( \frac{T}{P} \right)^\gamma = \text{const.}$$

$$\frac{T^\gamma}{P^{\gamma-1}} = \text{const.}$$

Curve (A) is isothermal

$T = \text{const.}$

$PV = \text{const.}$

37. B

**Sol.** Magnetic moment =  $i\pi r^2$

$$\mu = \frac{evr}{2}$$

$$\mu \propto \left(\frac{1}{n}\right) n^2$$

$\mu \propto n$

$x = 1$

38. B

**Sol.** Rising half to peak

$$t = T/6$$

$$t = \frac{2\pi}{6\omega} = \frac{\pi}{3\omega} = \frac{\pi}{300\pi} = \frac{1}{300} = 3.33 \text{ ms}$$

39. A

$$\text{Sol. } \frac{dy}{dx} = \tan \theta = \frac{x}{2} = \mu = \frac{1}{2}$$

$$x = 1, y = 1/4$$

40. B

$$\text{Sol. } p = \frac{E}{C} = \frac{6.48 \times 10^5}{3 \times 10^8} = 2.16 \times 10^{-3}$$

41. A

**Sol.** Intensity of emergent light

$$= \frac{I_0}{2} \cos^2 45^\circ = \frac{I_0}{4}$$

42. D

$$\text{Sol. } R_p = \frac{R_e}{3}, M_p = \frac{M_e}{6}$$

$$V_e = \sqrt{\frac{2GM_e}{R_e}} \quad \dots \dots \dots \text{(i)}$$

$$V_p = \sqrt{\frac{2GM_p}{R_p}} \quad \dots \dots \dots \text{(ii)}$$

$$\frac{V_e}{V_p} = \sqrt{2}$$

$$V_p = \frac{V_e}{\sqrt{2}} = \frac{11.2}{\sqrt{2}} = 7.9 \text{ km/sec}$$

43. B

$$\text{Sol. } \frac{2k\lambda q}{r} = m\omega^2 r$$

$$\omega^2 = \frac{2k\lambda q}{mr^2}$$

$$\left(\frac{2\pi}{T}\right)^2 = \frac{2k\lambda q}{mr^2}$$

$$T = 2\pi \sqrt{\frac{m}{2k\lambda q}}$$

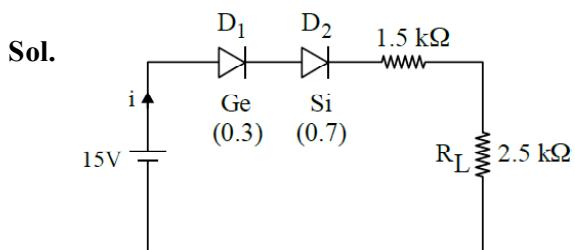
44. A

$$\text{Sol. } m = k c^p G^{-1/2} h^{1/2}$$

$$M^1 L^0 T^0 = [L T^{-1}]^p [M^{-1} L^3 T^{-2}]^{-1/2} [M L^2 T^{-1}]^{1/2}$$

By comparing  $P = 1/2$

45. A



$$i = \frac{14}{4} = 3.5 \text{ mA}$$

$$V_L = i R_L = 3.5 \times 2.5 \text{ volt} \\ = 8.75 \text{ volt}$$

46. C

$$\text{Sol. } f_1 = 3, f_2 = 5$$

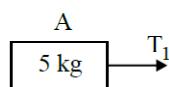
$$n_1 = 3, n_2 = 2$$

$$f_{\text{mixture}} = \frac{n_1 f_1 + n_2 f_2}{n_1 + n_2} = \frac{9 + 10}{5} = \frac{19}{5}$$

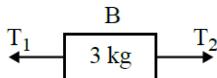
$$\gamma_{\text{mixture}} = 1 + \frac{2 \times 5}{19} = \frac{29}{19} = 1.52$$

47. A

$$\text{Sol. } a_A = a_B = a_C = \frac{F}{5+3+2} = \frac{80}{10} = 8 \text{ m/s}^2$$



$$T_1 = 5 \times 8 = 40$$



$$T_2 - T_1 = 3 \times 8 \Rightarrow T_2 = 64$$

48. D

**Sol.**  $\frac{v^2}{R} = W$  .....(i)

$$\frac{v^2}{\frac{1}{2}\left(\frac{R}{2}\right)} = W' \quad \dots\dots\text{(ii)}$$

From (i) & (ii), we get

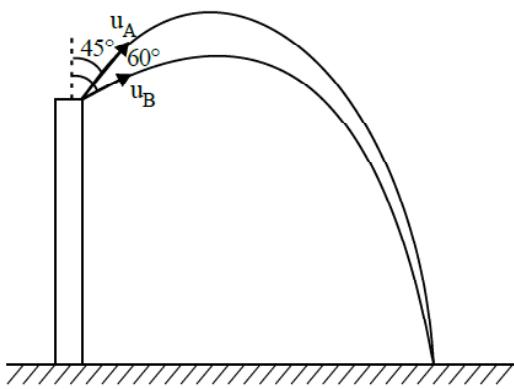
$$W' = 4W$$

49. D

**Sol.** Maxwell's equation

50. Bonus

**Sol.**



For  $u_A$  &  $u_B$  time of flight and range can not be same. So above options are incorrect.

### Section - B (Numerical Value Type)

51. 45

**Sol.**  $P_i = 2300 \times 5 \text{ watt}$

$$P_0 = 2300 \times 5 \times 0.9 = 230 \times I_2$$

$$I_2 = 45A$$

52. 10

**Sol.**  $\rho \left( \frac{4}{3} \pi r^3 \right) 1000 = \frac{4}{3} \pi R^3 \rho$

$$R = 10r$$

$$E_1 = 1000 \times 4\pi r^2 \times S$$

$$E_2 = 4\pi (10r)^2 S$$

$$\frac{E_1}{E_2} = \frac{10}{1}, x = 10$$

53. 24

**Sol.**  $I_1 \omega_1 + I_2 \omega_2 = (I_1 + I_2) \omega_0$  (C.O.A.M.)

gives  $\omega_0 = 8 \text{ rad/s}$

$$E_1 = \frac{1}{2} I_1 \omega_1^2 + \frac{1}{2} I_2 \omega_2^2 = 216 \text{ J}$$

$$E_2 = \frac{1}{2} (I_1 + I_2) \omega_0^2 = 192 \text{ J}$$

$$\therefore \Delta E = 24 \text{ J}$$

54. 20

**Sol.**  $\frac{1}{f+20} - \frac{1}{-(f+20)} = \frac{1}{f}$

$$\frac{2}{f+20} = \frac{1}{f} \quad f = 20 \text{ cm}$$

$$\text{Or } x_1 x_2 = f^2 \text{ gives } f = 20 \text{ cm}$$

55. 4

**Sol.**  $\bar{N} = |\bar{A}| \hat{B} = \frac{5(4\hat{i} + 3\hat{j})}{5} = 4\hat{i} + 3\hat{j}$

$$\therefore x = 4$$

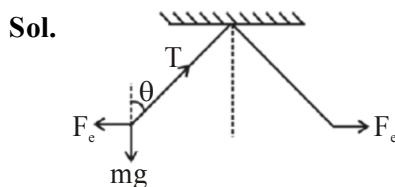
56. 40

**Sol.**  $B = 4 \times \frac{\mu_0 i}{4\pi \left( \frac{1}{2} \right)} \left( \frac{1}{\sqrt{2}} + \frac{1}{\sqrt{2}} \right)$

$$= 4 \times 10^{-7} \times 5 \times 2 \times \sqrt{2}$$

$$= 40\sqrt{2} \times 10^{-7} \text{ T}$$

57. 2



$$T \cos \theta = mg$$

$$T \sin \theta = F_e$$

$$\tan \theta = \frac{F_e}{mg}$$

$$\tan \theta = \frac{F_e}{\rho_B V g} \quad \dots\dots\text{(i)}$$

$$\tan \theta = \frac{F_e}{\frac{k}{(\rho_B - \rho_L) V g}} \quad \dots\dots\text{(ii)}$$

From Eq. (i) & (ii)

$$\rho_B Vg = (\rho_B - \rho_L) k Vg$$

$$1.4 = 0.7 k$$

$$k = 2$$

58. 8

**Sol.** Acceleration due to gravity  $g' = \frac{g}{4}$

$$T = 2\pi \sqrt{\frac{4l}{g}}$$

$$T = 2\pi \sqrt{\frac{4 \times 4}{g}}$$

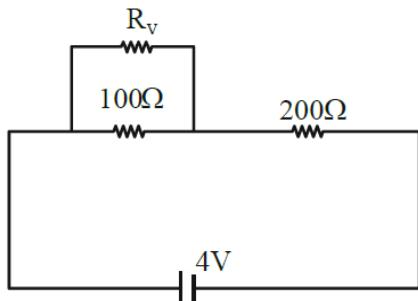
$$T = 2\pi \frac{4}{\pi} = 8s$$

59. Bonus

**Sol.** Question is wrong as data is incomplete.

60. 200

**Sol.**



$$\frac{R_v 100}{R_v + 100} = \frac{200}{3}$$

$$3R_v = 2R_v + 200$$

$$R_v = 200$$

## CHEMISTRY

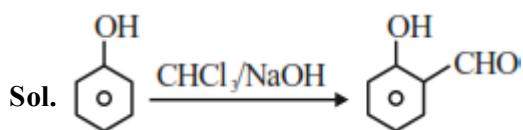
### Section - A (Single Correct Answer)

61. (D)

**Sol.** Different Extraction

Different layers are formed which can be separated in funnel. (Theory based)

62. (D)



63. (D)

**Sol.** Statement-I : Rate of  $S_N2$   $\propto [R-X] [Nu^-]$

$S_N2$  reaction is favoured by high concentration of nucleophile ( $Nu^-$ ) and less crowding in the substrate molecule.

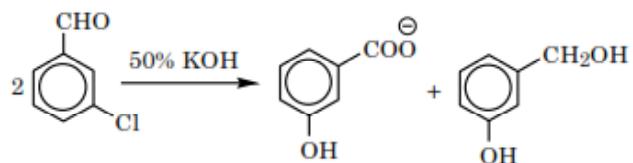
Statement-II : Solvolysis follows  $S_N1$  path.

Both are correct Statements.

64. (B)

**Sol.** Meta-chlorobenzaldehyde will undergo

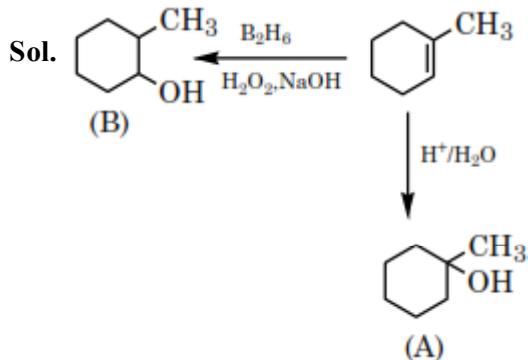
Cannizzaro reaction with 50% KOH to give m-chlorobenzoate ion and m-chlorobenzyl alcohol.



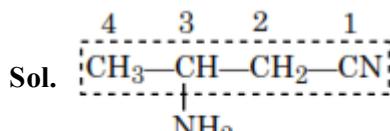
65. (B)

**Sol.** Due to lower Bond dissociation enthalpy of  $H_2Te$  it ionizes to give  $H^+$  more easily than  $H_2S$ .

66. (B)

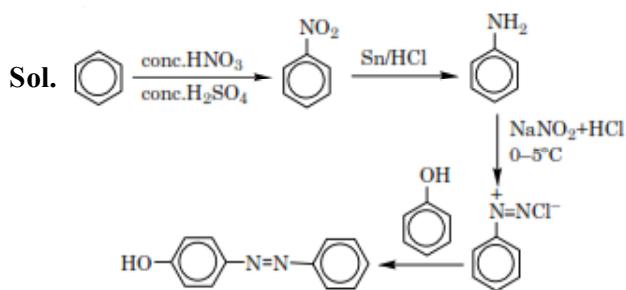


67. (C)



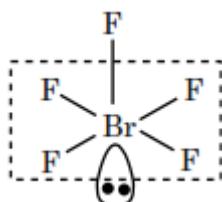
3-Aminobutanenitrile

68. (C)



69. (C)

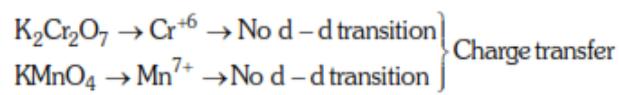
**Sol.**  $\text{BrF}_5$



**Square Pyramidal.**

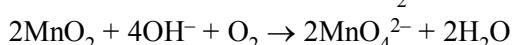
70. (A)

**Sol.**

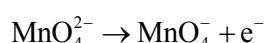


71. (B)

**Sol.** Alkaline oxidative fusion of  $\text{MnO}_2$ :



Electrolytic oxidation of  $\text{MnO}_4^{2-}$  in alkaline medium.



72. (B)

**Sol.** Mole fraction of C =  $\frac{n_C}{n_A + n_B + n_C}$

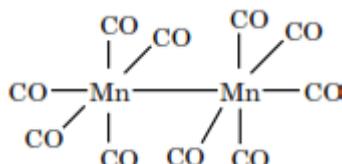
73. (C)

**Sol.** Chemical reactivity of elements decreases along the period therefore statement -I is false.

**Group – 1** elements from basic nature oxides while group – 17 elements form acidic oxides therefore statement-II is true.

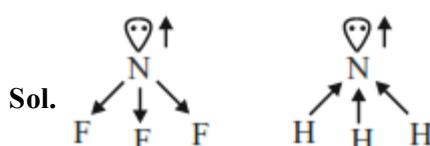
74. (A)

**Sol.**  $\text{Mn}_2(\text{CO})_{10}$



Octahedral around Mn

75. (B)



76. (C)

**Sol.** More number of hyperconjugable Hydrogens, more stable is the carbocations.

77. (A)

**Sol.**  $\Delta T_f$  is maximum when  $i \times m$  is maximum.

$$(1) m_1 = \frac{180}{60} = 3, i = 1 + \alpha$$

Hence

$$\Delta T_f = (1 + \alpha) \cdot k_f = 3 \times 1.86 = 5.58 \text{ }^\circ\text{C} (\alpha \ll 1)$$

$$(2) m_2 = \frac{180}{60} = 3, i = 0.5, \Delta T_f = \frac{3}{2} \times k_f' = 7.68 \text{ }^\circ\text{C}$$

$$(3) m_3 = \frac{180}{122} = 1.48, i = 0.5, \Delta T_f = \frac{1.48}{2} \times k_f' = 3.8 \text{ }^\circ\text{C}$$

$$(4) m_4 = \frac{180}{180} = 1, i = 1, \Delta T_f = 1 \cdot k_f' = 1.86 \text{ }^\circ\text{C}$$

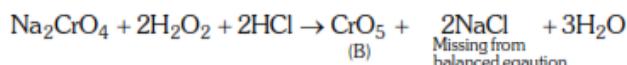
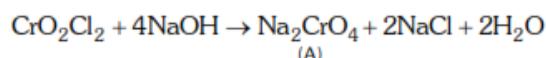
As per NCERT,

$$k_f' (\text{H}_2\text{O}) = 1.86 \text{ k} \cdot \text{kg mol}^{-1}$$

$$k_f' (\text{Benzene}) = 5.12 \text{ k} \cdot \text{kg mol}^{-1}$$

78. (A)

**Sol.**



(B) Missing from balanced equation

79. (C)

**Sol.** On moving down the group, bond strength of M–H bond decreases, which reduces the thermal stability but increases reducing nature of hydrides, hence A and B are correct statements.

80. (B)

**Sol.** Higher the value of  $\oplus$ ve SRP (Std. reduction potential) more is tendency to undergo reduction, so better is oxidising power of reactant.

Hence, ox. Power :  $\text{BrO}_4^- > \text{IO}_4^- > \text{ClO}_4^-$

### Section - B (Numerical Value Type)

81. (4)

**Sol.** cis –  $[\text{Cr}(\text{ox})_2\text{Cl}_2]^{3-}$  → can show optical isomerism (no POS & COS)

$[\text{Co}(\text{en})_3]^{3+}$  → can show (no POS & COS)

cis –  $[\text{Pt}(\text{en})_2\text{Cl}_2]^{2+}$  → can show (no POS & COS)

cis –  $[\text{Co}(\text{en})_2\text{Cl}_2]^+$  → can show (no POS & COS)

trans –  $[\text{Pt}(\text{en})_2\text{Cl}_2]^{2+}$  → can't show (contains POS & COS)

trans –  $[\text{Cr}(\text{ox})_2\text{Cl}_2]^{3-}$  → can't show (contains POS & COS)

82. (17)

**Sol.** Rate of reaction (ROR)

$$= -\frac{1}{2} \frac{\Delta [\text{N}_2\text{O}_5]}{\Delta t} = \frac{1}{4} \frac{[\text{NO}_2]}{\Delta t} = \frac{\Delta [\text{O}_2]}{\Delta t}$$

$$\text{ROR} = -\frac{1}{2} \frac{\Delta [\text{N}_2\text{O}_5]}{\Delta t} = -\frac{1}{2} \frac{(2.75 - 3)}{30} \text{ mol L}^{-1} \text{ min}^{-1}$$

$$\text{ROR} = -\frac{1}{2} \frac{(-0.25)}{30} \text{ mol L}^{-1} \text{ min}^{-1}$$

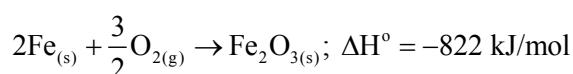
$$\text{ROR} = \frac{1}{240} \text{ mol L}^{-1} \text{ min}^{-1}$$

$$\text{Rate of formation } \text{NO}_2 = \frac{\Delta [\text{NO}_2]}{\Delta t} = 4 \times \text{ROR}$$

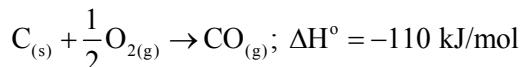
$$= \frac{4}{240} = 16.66 \times 10^{-3} \text{ mol L}^{-1} \text{ min}^{-1} \approx 17 \times 10^{-3}$$

83. (492)

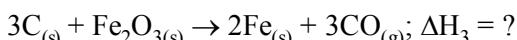
**Sol.**



..... (1)



..... (2)



(3) =  $3 \times (2) - (1)$

$$\Delta H_3 = 3 \times \Delta H_2 - \Delta H_1$$

$$= 3(-110) + 822$$

$$= 492 \text{ kJ/mole}$$

84. (3)

- Sol.** A. RNA is regarded as the reserve of genetic information. (False)
- B. DNA molecule self-duplicates during cell division. (True)
- C. DNA synthesizes proteins in the cell. (False)
- D. The message for the synthesis of particular proteins is present in DNA. (True)
- E. Identical DNA strands are transferred to daughter cells. (True)

85. (100)

**Sol.**  $1\text{M Benzoic acid} + 1\text{M Sodium Benzoate}$   
 $(V_a \text{ ml}) \quad (V_s \text{ ml})$   
Millimole  $V_a \times 1 \quad V_s \times 1$

$$\text{pH} = 4.5$$

$$\text{pH} = \text{pKa} + \log \frac{[\text{salt}]}{[\text{acid}]}$$

$$4.5 = 4.2 + \log \left( \frac{V_s}{V_a} \right)$$

$$\frac{V_s}{V_a} = 2 \quad \dots\dots (1)$$

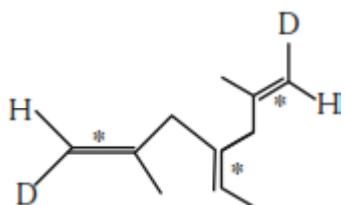
$$V_s + V_a = 300 \quad \dots\dots (2)$$

$$V_a = 100 \text{ ml}$$

86. (4)

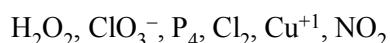
**Sol.** 3 stereocenters, symmetrical

Total Geometrical isomers  $\rightarrow$  4. EE, ZZ, EZ (two isomers)



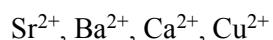
87. (6)

**Sol.** Intermediate oxidation state of element can undergo disproportionation.

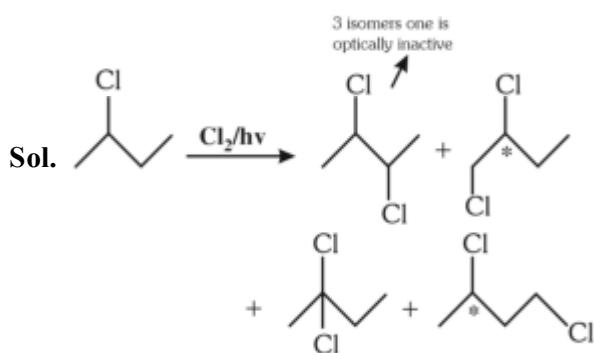


88. (4)

**Sol.** All the following metal ions will respond to flame test.



89. (6)



90. (10)

**Sol.** 5<sup>th</sup> excited state  $\Rightarrow n_1 = 6$

1<sup>st</sup> excited state  $\Rightarrow n_2 = 2$

$$\Delta n = n_1 - n_2 = 6 - 2 = 4$$

Maximum number of spectral lines

$$= \frac{\Delta n(\Delta n + 1)}{2} = \frac{4(4 + 1)}{2} = 10$$

● ● ●