

MATHEMATICS Section - A (Single Correct Answer)

- ${}^{n-1}C = (k^2 - 8) {}^nC_{r+1}$ if and only if :

(A) $2\sqrt{2} < k \leq 3$ (B) $2\sqrt{3} < k \leq 3\sqrt{2}$

(C) $2\sqrt{3} < k < 3\sqrt{3}$ (D) $2\sqrt{2} < k < 2\sqrt{3}$
- The distance, of the point $(7, -2, 11)$ from the line $\frac{x-6}{1} = \frac{y-4}{0} = \frac{z-8}{3}$ along the line $\frac{x-5}{2} = \frac{y-1}{-3} = \frac{z-5}{6}$, is :

(A) 12 (B) 14 (C) 18 (D) 21
- Let $x = x(t)$ and $y = y(t)$ be solutions of the differential equations $\frac{dx}{dt} + ax = 0$ and $\frac{dy}{dt} + by = 0$ respectively, $a, b \in \mathbb{R}$. Given that $x(0) = 2$; $y(0) = 1$ and $3y(1) = 2x(1)$, the value of t , for which $x(t) = y(t)$, is :

(A) $\log_{\frac{2}{3}} 2$ (B) $\log_4 3$

(C) $\log_3 4$ (D) $\log_{\frac{4}{5}} 2$
- If (a, b) be the orthocentre of the triangle whose vertices are $(1, 2)$, $(2, 3)$ and $(3, 1)$ and $I_1 = \int_a^b x \sin(4x - x^2) dx$, $I_2 = \int_a^b \sin(4x - x^2) dx$, then $36 \frac{I_1}{I_2}$ is equal to :

(A) 72 (B) 88 (C) 80 (D) 66
- If A denotes the sum of all the coefficients in the expansion of $(1 - 3x + 10x^2)^n$ and B denotes the sum of all the coefficients in the expansion of $(1 + x^2)^n$, then :

(A) $A = B^3$ (B) $3A = B$

(C) $B = A^3$ (D) $A = 3B$
- The number of common terms in the progressions $4, 9, 14, 19, \dots$, up to 25^{th} term and $3, 6, 9, 12, \dots$, up to 37^{th} term is :

(A) 9 (B) 5 (C) 7 (D) 8
- If the shortest distance of the parabola $y^2 = 4x$ from the centre of the circle $x^2 + y^2 - 4x - 16y + 64 = 0$ is d , then d^2 is equal to :

(A) 16 (B) 24 (C) 20 (D) 36
- If the shortest distance between the lines $\frac{x-4}{1} = \frac{y-1}{2} = \frac{z}{-3}$ and $\frac{x-\lambda}{2} = \frac{y+1}{4} = \frac{z-2}{-5}$ is $\frac{6}{\sqrt{5}}$, then the sum of all possible values of λ is :

(A) 5 (B) 8 (C) 7 (D) 10

9. If $\int_0^1 \frac{1}{\sqrt{3+x} + \sqrt{1+x}} dx = a + b\sqrt{2} + c\sqrt{3}$, where a, b, c are rational numbers, then $2a + 3b - 4c$ is equal to
 (A) 4 (B) 10 (C) 7 (D) 8
10. Let $S = \{1, 2, 3, \dots, 10\}$. Suppose M is the set of all the subsets of S , then the relation $R = \{(A, B) : A \cap B \neq \phi; A, B \in M\}$ is :
 (A) symmetric and reflexive only (B) reflexive only
 (C) symmetric and transitive only (D) symmetric only
11. If $S = \{z \in \mathbb{C} : |z - i| = |z + i| = |z - 1|\}$, then, $n(S)$ is :
 (A) 1 (B) 0 (C) 3 (D) 2
12. Four distinct points $(2k, 3k), (1, 0), (0, 1)$ and $(0, 0)$ lie on a circle for k equal to :
 (A) $\frac{2}{13}$ (B) $\frac{3}{13}$ (C) $\frac{5}{13}$ (D) $\frac{1}{13}$

13. Consider the function, $f(x) = \begin{cases} \frac{(7x - 12 - x^2)}{b|x^2 - 7x + 12|} & , x < 3 \\ 2^{\frac{\sin(x-3)}{x-[x]}} & , x > 3 \\ b & , x = 3 \end{cases}$

Where $[x]$ denotes the greatest integer less than or equal to x . If S denotes the set of all ordered pairs (a, b) such that $f(x)$ is continuous at $x = 3$, then the number of elements in S is :

- (A) 2 (B) Infinitely many (C) 4 (D) 1
14. Let a_1, a_2, \dots, a_{10} be 10 observations such that $\sum_{k=1}^{10} a_k = 50$ and $\sum_{\forall k < j} a_k \cdot a_j = 1100$. Then the standard deviation of a_1, a_2, \dots, a_{10} is equal to :
 (A) 5 (B) $\sqrt{5}$ (C) 10 (D) $\sqrt{115}$
15. The length of the chord of the ellipse $\frac{x^2}{25} + \frac{y^2}{16} = 1$, whose mid point is $\left(1, \frac{2}{5}\right)$, is equal to :
 (A) $\frac{\sqrt{1691}}{5}$ (B) $\frac{\sqrt{2009}}{5}$ (C) $\frac{\sqrt{1741}}{5}$ (D) $\frac{\sqrt{1541}}{5}$
16. The portion of the line $4x + 5y = 20$ in the first quadrant is trisected by the lines L_1 and L_2 passing through the origin. The tangent of an angle between the lines L_1 and L_2 is :
 (A) $\frac{8}{5}$ (B) $\frac{25}{41}$ (C) $\frac{2}{5}$ (D) $\frac{30}{41}$
17. Let $\vec{a} = \hat{i} + 2\hat{j} + k, \vec{b} = 3(\hat{i} - \hat{j} + k)$. Let \vec{c} be the vector such that $\vec{a} \times \vec{c} = \vec{b}$ and $\vec{a} \cdot \vec{c} = 3$. Then $\vec{a} \cdot ((\vec{c} \times \vec{b}) - \vec{b} - \vec{c})$ is equal to :
 (A) 32 (B) 24 (C) 20 (D) 36

18. If $a = \lim_{x \rightarrow 0} \frac{\sqrt{1 + \sqrt{1 + x^4}} - \sqrt{2}}{x^4}$ and $b = \lim_{x \rightarrow 0} \frac{\sin^2 x}{\sqrt{2} - \sqrt{1 + \cos x}}$, then the value of ab^3 is :
- (A) 36 (B) 32 (C) 25 (D) 30

19. Consider the matrix $f(x) = \begin{bmatrix} \cos x & -\sin x & 0 \\ \sin x & \cos x & 0 \\ 0 & 0 & 1 \end{bmatrix}$.

Given below are two statements :

Statement I : $f(-x)$ is the inverse of the matrix $f(x)$.

Statement II : $f(x) f(y) = f(x + y)$.

In the light of the above statements, choose the correct answer from the options given below

- (A) Statement I is false but Statement II is true
 (B) Both Statement I and Statement II are false
 (C) Statement I is true but Statement II is false
 (D) Both Statement I and Statement II are true
20. The function $f : \mathbb{N} - \{1\} \rightarrow \mathbb{N}$; defined by $f(n) =$ the highest prime factor of n , is :
- (A) both one-one and onto (B) one-one only
 (C) onto only (D) neither one-one nor onto

Section - B (Numerical Value Type)

21. The least positive integral value of α , for which the angle between the vectors $\alpha \hat{i} - 2\hat{j} + 2\hat{k}$ and $\alpha \hat{i} + 2\hat{j} - 2\hat{k}$ is acute, is _____
22. Let for a differentiable function $f : (0, \infty) \rightarrow \mathbb{R}$, $f(x) - f(y) \geq \log_e \left(\frac{x}{y} \right) + x - y$, $\forall x, y \in (0, \infty)$.
 Then $\sum_{n=1}^{20} f' \left(\frac{1}{n^2} \right)$ is equal to _____.
23. If the solution of the differential equation $(2x + 3y - 2) dx + (4x + 6y - 7) dy = 0$, $y(0) = 3$, is $\alpha x + \beta y + 3 \log_e |2x + 3y - \gamma| = 6$, then $\alpha + 2\beta + 3\gamma$ is equal to _____.
24. Let the area of the region $\{(x, y) : x - 2y + 4 \geq 0, x + 2y^2 \geq 0, x + 4y^2 \leq 8, y \geq 0\}$ be $\frac{m}{n}$, where m and n are coprime numbers. Then $m + n$ is equal to _____.
25. If $8 = 3 + \frac{1}{4}(3+p) + \frac{1}{4^2}(3+2p) + \frac{1}{4^3}(3+3p) + \dots \infty$, then the value of p is _____.
26. A fair die is tossed repeatedly until a six is obtained. Let X denote the number of tosses required and let $a = P(X = 3)$, $b = P(X \geq 3)$ and $c = P(X \geq 6 | X > 3)$.
 Then $\frac{b+c}{a}$ is equal to _____.

27. Let the set of all $a \in \mathbb{R}$ such that the equation $\cos 2x + a \sin x = 2a - 7$ has a solution be $[p, q]$ and $r = \tan 9^\circ - \tan 27^\circ - \frac{1}{\cot 63^\circ} + \tan 81^\circ$, then pqr is equal to _____.
28. Let $f(x) = x^3 + x^2f'(1) + xf''(2) + f'''(3)$, $x \in \mathbb{R}$. Then $f'(10)$ is equal to _____.
29. Let $A = \begin{bmatrix} 2 & 0 & 1 \\ 1 & 1 & 0 \\ 1 & 0 & 1 \end{bmatrix}$, $B = [B_1, B_2, B_3]$, where B_1, B_2, B_3 are column matrices, and
- $$AB_1 = \begin{bmatrix} 1 \\ 0 \\ 0 \end{bmatrix}, AB_2 = \begin{bmatrix} 2 \\ 3 \\ 0 \end{bmatrix}, AB_3 = \begin{bmatrix} 3 \\ 2 \\ 1 \end{bmatrix}$$
- If $\alpha = |B|$ and β is the sum of all the diagonal elements of B , then $\alpha^3 + \beta^3$ is equal to _____.
30. If α satisfies the equation $x^2 + x + 1 = 0$ and $(1 + \alpha)^7 = A + B\alpha + C\alpha^2$, $A, B, C \geq 0$, then $5(3A - 2B - C)$ is equal to _____.

PHYSICS

Section - A (Single Correct Answer)

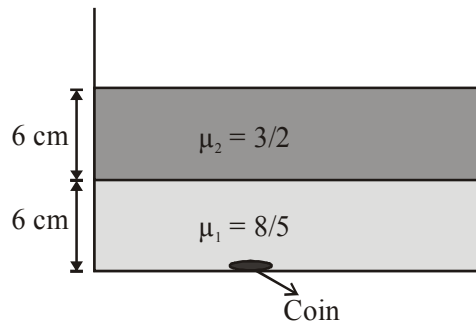
31. Position of an ant (S in metres) moving in Y - Z plane is given by $S = 2t^2\hat{j} + 5t\hat{k}$. (where t is in second). The magnitude and direction of velocity of the ant at $t = 1$ s will be :
- (A) 16 m/s in y -direction (B) 4 m/s in x -direction
 (C) 9 m/s in z -direction (D) 4 m/s in y -direction
32. Given below are two statements :
- Statement (I)** : Viscosity of gases is greater than that of liquids.
Statement (II) : Surface tension of a liquid decreases due to the presence of insoluble impurities.
- In the light of the above statements, choose the most appropriate answer from the options given below :
- (A) Statement I is correct but statement II is incorrect
 (B) Statement I is incorrect but Statement II is correct
 (C) Both Statement I and Statement II are incorrect
 (D) Both Statement I and Statement II are correct
33. If the refractive index of the material of a prism is $\cot\left(\frac{A}{2}\right)$, where A is the angle of prism then the angle of minimum deviation will be
- (A) $\pi - 2A$ (B) $\frac{\pi}{2} - 2A$ (C) $\pi - A$ (D) $\frac{\pi}{2} - A$
34. A proton moving with a constant velocity passes through a region of space without any change in its velocity. If \vec{E} and \vec{B} represent the electric and magnetic fields respectively, then the region of space may have :
- (1) $E = 0, B = 0$ (2) $E = 0, B \neq 0$
 (3) $E \neq 0, B = 0$ (4) $E \neq 0, B \neq 0$

44. A convex lens of focal length 40 cm forms an image of an extended source of light on a photoelectric cell. A current I is produced. The lens is replaced by another convex lens having the same diameter but focal length 20 cm. The photoelectric current now is :
- (A) $I/2$ (B) $4I$ (C) $2I$ (D) I
45. A body of mass 1000 kg is moving horizontally with a velocity 6 m/s. If 200 kg extra mass is added, the final velocity (in m/s) is:
- (A) 6 (B) 2 (C) 3 (D) 5
46. A plane electromagnetic wave propagating in x-direction is described by $E_y = (200 \text{ Vm}^{-1}) \sin[1.5 \times 10^7 t - 0.05 x]$; The intensity of the wave is :
- (Use $\epsilon_0 = 8.85 \times 10^{-12} \text{ C}^2\text{N}^{-1} \text{ m}^{-2}$)
- (A) 35.4 Wm^{-2} (B) 53.1 Wm^{-2} (C) 26.6 Wm^{-2} (D) 106.2 Wm^{-2}
47. Given below are two statements :
- Statement (I)** : Planck's constant and angular momentum have same dimensions.
Statement (II) : Linear momentum and moment of force have same dimensions.
 In the light of the above statements, choose the correct answer from the options given below :
- (A) Statement I is true but Statement II is false
 (B) Both Statement I and Statement II are false
 (C) Both Statement I and Statement II are true
 (D) Statement I is false but Statement II is true
48. A wire of length 10 cm and radius $\sqrt{7} \times 10^{-4} \text{ m}$ connected across the right gap of a meter bridge. When a resistance of 4.5Ω is connected on the left gap by using a resistance box, the balance length is found to be at 60 cm from the left end. If the resistivity of the wire is $R \times 10^{-7} \Omega\text{m}$, then value of R is :
- (A) 63 (B) 70 (C) 66 (D) 35
49. A wire of resistance R and length L is cut into 5 equal parts. If these parts are joined parallelly, then resultant resistance will be :
- (A) $\frac{1}{25}R$ (B) $\frac{1}{5}R$ (C) $25R$ (D) $5R$
50. The average kinetic energy of a monatomic molecule is 0.414 eV at temperature :
- (Use $K_B = 1.38 \times 10^{-23} \text{ J/mol-K}$)
- (A) 3000 K (B) 3200 K (C) 1600 K (D) 1500 K

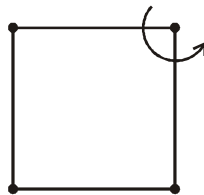
Section - B (Numerical Value Type)

51. A particle starts from origin at $t = 0$ with a velocity $5\hat{i}$ m/s and moves in x-y plane under action of a force which produces a constant acceleration of $(3\hat{i} + 2\hat{j})\text{m/s}^2$. If the x-coordinate of the particle at that instant is 84 m, then the speed of the particle at this time is $\sqrt{\alpha}$ m / s. The value of α is _____.
52. A thin metallic wire having cross sectional area of 10^{-4} m^2 is used to make a ring of radius 30 cm. A positive charge of $2\pi \text{ C}$ is uniformly distributed over the ring, while another positive charge of 30 pC is kept at the centre of the ring. The tension in the ring is _____ N ; provided that the ring does not get deformed.
 (neglect the influence of gravity). (given, $\frac{1}{4\pi\epsilon_0} = 9 \times 10^9 \text{ SI units}$)

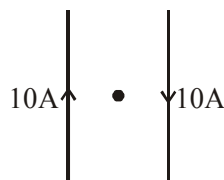
53. Two coils have mutual inductance 0.002 H. The current changes in the first coil according to the relation $i = i_0 \sin \omega t$, where $i_0 = 5\text{A}$ and $\omega = 50\pi \text{ rad/s}$. The maximum value of emf in the second coil is $\frac{\pi}{\alpha} \text{V}$. The value of α is _____.
54. Two immiscible liquids of refractive indices $8/5$ and $3/2$ respectively are put in a beaker as shown in the figure. The height of each column is 6 cm. A coin is placed at the bottom of the beaker. For near normal vision, the apparent depth of the coin is $\frac{\alpha}{4} \text{cm}$. The value of α is _____.



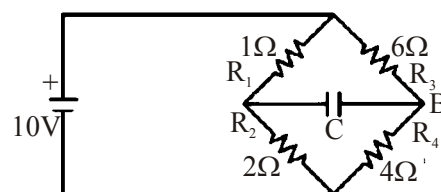
55. In a nuclear fission process, a high mass nuclide ($A \approx 236$) with binding energy 7.6 MeV/Nucleon dissociated into middle mass nuclides ($A \approx 118$), having binding energy of 8.6 MeV/Nucleon. The energy released in the process would be _____ MeV.
56. Four particles each of mass 1 kg are placed at four corners of a square of side 2 m. Moment of inertia of system about an axis perpendicular to its plane and passing through one of its vertex is _____ kgm^2 .



57. A particle executes simple harmonic motion with an amplitude of 4 cm. At the mean position, velocity of the particle is 10 cm/s. The distance of the particle from the mean position when its speed becomes 5 cm/s is $\sqrt{\alpha} \text{cm}$, where $\alpha =$ _____.
58. Two long, straight wires carry equal currents in opposite directions as shown in figure. The separation between the wires is 5.0 cm. The magnitude of the magnetic field at a point P midway between the wires is _____ μT (Given : $\mu_0 = 4\pi \times 10^{-7} \text{ TmA}^{-1}$)



59. The charge accumulated on the capacitor connected in the following circuit is _____ μC (Given $C = 150 \mu\text{F}$)



60. If average depth of an ocean is 4000 m and the bulk modulus of water is $2 \times 10^9 \text{ Nm}^{-2}$, then fractional compression $\frac{\Delta V}{V}$ of water at the bottom of ocean is $\alpha \times 10^{-2}$. The value of α is _____
(Given, $g = 10 \text{ ms}^{-2}$, $\rho = 1000 \text{ kg m}^{-3}$).

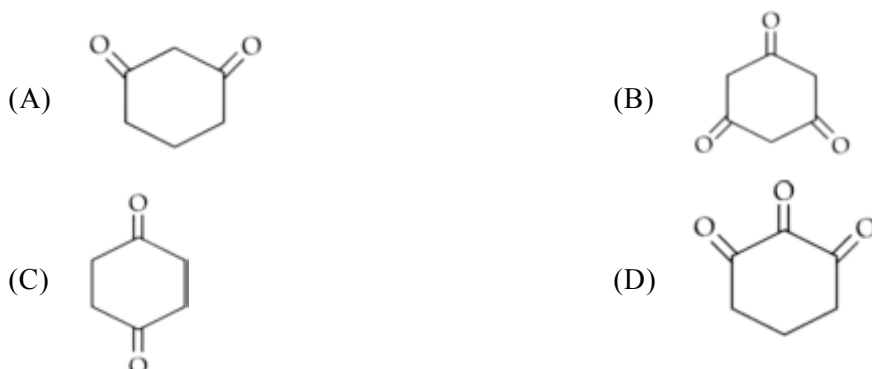
CHEMISTRY

Section - A (Single Correct Answer)

61. Two nucleotides are joined together by a linkage known as :

- (A) Phosphodiester linkage (B) Glycosidic linkage
(C) Disulphide linkage (D) Peptide linkage

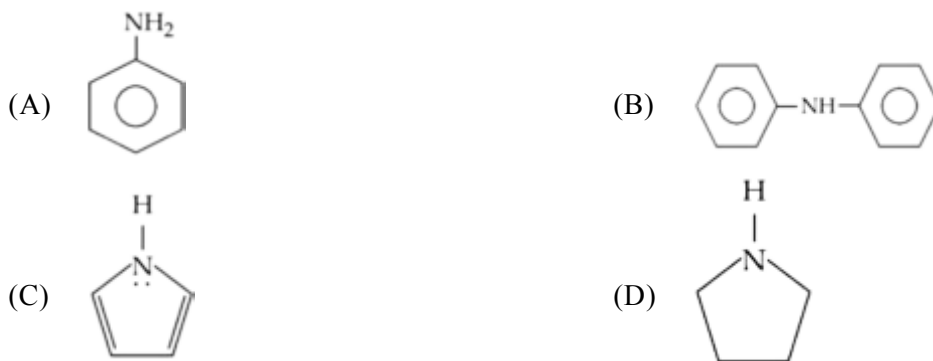
62. Highest enol content will be shown by



63. Element not showing variable oxidation state is

- (A) Bromine (B) Iodine (C) Chlorine (D) Fluorine

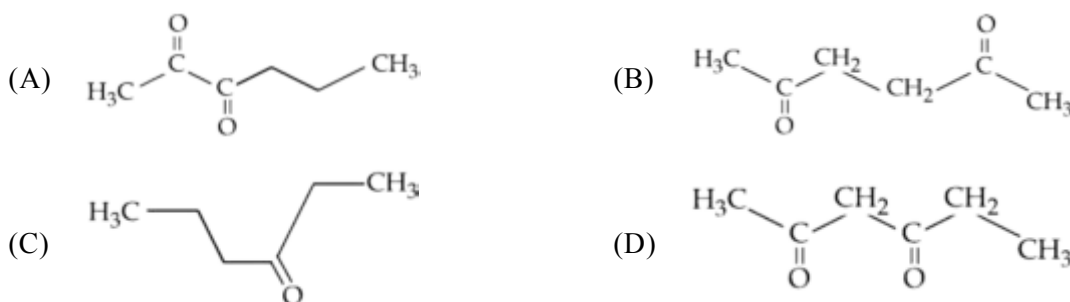
64. Which of the following is strongest Bronsted base ?



65. Which of the following electronic configuration would be associated with the highest magnetic moment ?

- (A) $[\text{Ar}] 3d^7$ (B) $[\text{Ar}] 3d^8$ (C) $[\text{Ar}] 3d^3$ (D) $[\text{Ar}] 3d^6$

66. Which of the following has highly acidic hydrogen ?



67. A solution of two miscible liquids showing negative deviation from Raoult's law will have :
- (A) increased vapour pressure, increased boiling point
 (B) increased vapour pressure, decreased boiling point
 (C) decreased vapour pressure, decreased boiling point
 (D) decreased vapour pressure, increased boiling point

68. Consider the following complex ions –



The correct order of the complex ions, according to their spin only magnetic moment values (in B.M.) is

- (A) $R < Q < P$ (B) $R < P < Q$ (C) $Q < R < P$ (D) $Q < P < R$
69. Choose the polar molecule from the following :
- (A) CCl_4 (B) CO_2 (C) $\text{CH}_2 = \text{CH}_2$ (D) CHCl_3
70. Given below are two statements :

Statement (I) : The 4f and 5f-series of elements are placed separately in the Periodic table to preserve the principle of classification.

Statement (II) : S-block elements can be found in pure form in nature.

In the light of the above statements, choose the most appropriate answer from the options given below.

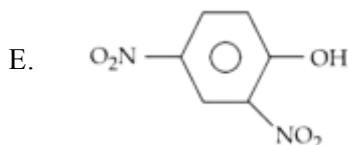
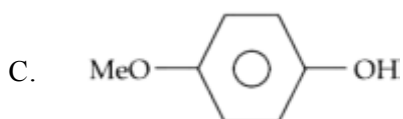
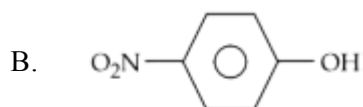
- (A) Statement I is false but Statement II is true
 (B) Both Statement I and Statement II are true
 (C) Statement I is true but Statement II is false
 (D) Both Statement I and Statement II are false
71. Given below are two statements :

Statement (I) : p-nitrophenol is more acidic than m-nitrophenol and o-nitrophenol.

Statement (II) : Ethanol will give immediate turbidity with Lucas reagent.

In the light of the above statements, choose the correct answer from the options given below.

- (A) Statement I is true but Statement II is false
 (B) Both Statement I and Statement II are true
 (C) Both Statement I and Statement II are false
 (D) Statement I is false but Statement II is true
72. The ascending order of acidity of –OH group in the following compounds is :



Choose the correct answer from the options given below.

- (A) $(A) < (D) < (C) < (B) < (E)$ (B) $(C) < (A) < (D) < (B) < (E)$
 (C) $(C) < (D) < (B) < (A) < (E)$ (D) $(A) < (C) < (D) < (B) < (E)$

73. Given below are two statements : one is labelled as Assertion (A) and the other is labelled as Reason (R).
Assertion (A) : Melting point of Boron (2453 K) is unusually high in group 13 elements.
Reason (R) : Solid Boron has very strong crystalline lattice.
 In the light of the above statements, choose the most appropriate answer from the options given below.
- (A) Both (A) and (R) are correct but (R) Is not the correct explanation of (A)
 (B) Both (A) and (R) are correct and (R) is the correct explanation of (A)
 (C) (A) is true but (R) is false
 (D) (A) is false but (R) is true

74. Cyclohexene  is _____ type of an organic compound.

- (A) Benzenoid aromatic (B) Benzenoid non-aromatic
 (C) Acyclic (D) Alicyclic
75. Yellow compound of lead chromate gets dissolved on treatment with hot NaOH solution. The product of lead formed is a :
- (A) Tetraanionic complex with coordination number six
 (B) Neutral complex with coordination number four
 (C) Dianionic complex with coordination number six
 (D) Dianionic complex with coordination number four

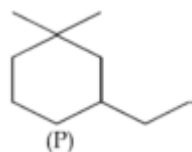
76. Given below are two statements :

Statement (I) : Aqueous solution of ammonium carbonate is basic.

Statement (II) : Acidic / basic nature of salt solution of a salt of weak acid and weak base depends on K_a and K_b value of acid and the base forming it.

In the light of the above statements, choose the most appropriate answer from the options given below.

- (A) Both Statement I and Statement II are correct
 (B) Statement I is correct but Statement II is incorrect
 (C) Both Statement I and Statement II are incorrect
 (D) Statement I is incorrect but Statement II is correct
77. IUPAC name of following compound (P) is :



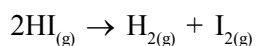
- (A) 1-Ethyl-5, 5-dimethylcyclohexane (B) 3-Ethyl-1, 1-dimethylcyclohexane
 (C) 1-Ethyl-3, 3-dimethylcyclohexane (D) 1,1-Dimethyl-3-ethylcyclohexane
78. NaCl reacts with conc. H_2SO_4 and $K_2Cr_2O_7$ to give reddish fumes (B), which react with NaOH to give yellow solution (C). (B) and (C) respectively are ;
- (A) CrO_2Cl_2 , Na_2CrO_4 (B) Na_2CrO_4 , CrO_2Cl_2
 (C) CrO_2Cl_2 , $KHSO_4$ (D) CrO_2Cl_2 , $Na_2Cr_2O_7$
79. The correct statement regarding nucleophilic substitution reaction in a chiral alkyl halide is ;
- (A) Retention occurs in S_N1 reaction and inversion occurs in S_N2 reaction.
 (B) Racemisation occurs in S_N1 reaction and retention occurs in S_N2 reaction.
 (C) Racemisation occurs in both S_N1 and S_N2 reactions.
 (D) Racemisation occurs in S_N1 reaction and inversion occurs in S_N2 reaction.

80. The electronic configuration for Neodymium is :
 [Atomic Number for Neodymium 60]
 (A) $[\text{Xe}] 4f^4 6s^2$ (B) $[\text{Xe}] 5f^4 7s^2$ (C) $[\text{Xe}] 4f^6 6s^2$ (D) $[\text{Xe}] 4f^1 5d^1 6s^2$

Section - B (Numerical Value Type)

81. The mass of silver (Molar mass of Ag : 108 g mol^{-1}) displaced by a quantity of electricity which displaces 5600 mL of O_2 at S.T.P. will be _____ g.

82. Consider the following data for the given reaction,



	1	2	3
HI (mol L^{-1})	0.005	0.01	0.02
Rate ($\text{mol L}^{-1} \text{ s}^{-1}$)	7.5×10^{-4}	3.0×10^{-3}	1.2×10^{-2}

The order of the reaction is _____.

83. Mass of methane required to produce 22 g of CO_2 after complete combustion is _____ g.

[Given : Molar mass in g mol^{-1}

$$\text{C} = 12.0$$

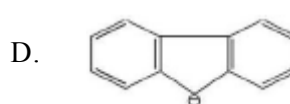
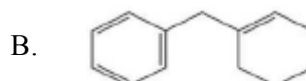
$$\text{H} = 1.0$$

$$\text{O} = 16.0]$$

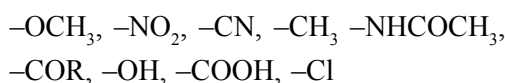
84. If three moles of an ideal gas at 300 K expand isothermally from 30 dm^3 to 45 dm^3 against a constant opposing pressure of 80 kPa, then the amount of heat transferred is _____ J.

85. 3-Methylhex-2-ene on reaction with HBr in presence of peroxide forms an addition product (A). The number of possible stereoisomers for 'A' is _____.

86. Among the given organic compounds, the total number of aromatic compounds is



87. Among the following, total number of meta directing functional groups is (Integer based)

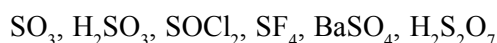


88. The number of electrons present in all the completely filled subshells having $n = 4$ and $s = +\frac{1}{2}$ is _____.

[Where, n = principal quantum number and s = spin quantum number]

89. Sum of bond order of CO and NO^+ is _____.

90. From the given list, the number of compounds with +4 oxidation state of Sulphur _____.



27-Jan.-2024 (Morning) : PCM

MATHEMATICS

Single Choice Correct					
1. A	2. B	3. D	4. A	5. A	
6. C	7. C	8. B	9. D	10. D	
11. D	12. C	13. D	14. B	15. A	
16. D	17. B	18. B	19. D	20. D	
Numerical Value					
21. 5	22. 2890	23. 29	24. 119	25. 9	
26. 12	27. 48	28. 202	29. 28	30. 5	

PHYSICS

Single Choice Correct					
31. D	32. B	33. A	34. C	35. D	
36. B	37. D	38. B	39. C	40. C	
41. B	42. C	43. C	44. D	45. D	
46. B	47. A	48. C	49. A	50. B	
Numerical Value					
51. 673	52. 3	53. 2	54. 31	55. 236	
56. 16	57. 12	58. 160	59. 400	60. 2	

CHEMISTRY

Single Choice Correct					
61. A	62. B	63. D	64. D	65. D	
66. D	67. D	68. C	69. D	70. C	
71. A	72. D	73. B	74. D	75. D	
76. A	77. B	78. A	79. D	80. A	
Numerical Value					
81. 107	82. 2	83. 8	84. 1200	85. 4	
86. 3	87. 4	88. 16	89. 6	90. 3	